

1 **Effect of Iron Ore and Copper Ore Tailings on Engineering Properties and Hydration**
2 **Products of Sustainable Cement Mortar**
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42 **Highlights**

- 43 • Fineness of mine tailings played a key role in increasing the performance
- 44 • The iron compounds in finer fraction of mine tailings accelerates the hydration
- 45 • Secondary hydration reactions were confirmed through advanced characterization.
- 46 • The formation of deleterious compounds restricted the high-volume sand replacement.

47 **Abstract**

48 The prohibition of river sand mining has convincingly drawn the research attention in finding
49 the practicable alternatives. In the approach of finding these alternatives, it is essential to ensure
50 a minimal or zero impairment to the ecological balance which can be mainly attained by
51 making use of industrial wastes/by-products. The wastes from the mining industries are the
52 major contributors in causing the impairment to environment, their influence on the stability
53 of mortars on using as fine aggregates need to be systematically investigated with the view of
54 long-term performance concerns.

55 Thus, the present study explores the applicability of mine tailings and finding the optimum
56 dosage in cement mortars by investigating the engineering properties and microstructure
57 development with the aid of qualitative and quantitative analysis associated to hydration
58 products. The studies confirm that, the increased consumption of portlandite for secondary
59 hydration reactions followed by the additional formation of calcium silicate hydrate (CSH) and
60 calcium aluminum silicate hydrate (CASH) phases in mine tailing-based mortars helped in
61 achieving a quality microstructure. These additional formations of CSH and CASH phases are
62 also confirmed through FTIR by identifying the shift of Si-O-Si stretching vibration bands
63 towards a lower wavenumber. The lowering of Ca/Si atomic ratio and increased formation of
64 mineralogical compounds related to CSH and CASH in XRD patterns also confirms the same.
65 Gismondine, Chabazite and Hillebrandite are the additional phases formed and found to be
66 taken part in refining the pore structure. This enhanced performance of mine tailing mortars
67 was also verified with the aid of modified Andreasen and Andersen particle packing model.
68 The formation of high-quality microstructure is reflected in the hardened properties of
69 optimized cement mortar which in the proportion of 20% for iron ore tailing (IOT) and 30%
70 for copper ore tailing (COT).

71 **Keywords:** Mine tailings, mortar, hydration, particle packing, microstructure, characterization,
72 sustainability.

73 1. Introduction

74 The ever-increasing quantities of industrial by-products and waste materials produced
75 worldwide, the solid waste management has become a major environmental concern.
76 Discharging industrial wastes such as mine tailings into the environment have a greater ability
77 to cause environmental hazards that can pose a substantial health risk to the creatures. In India,
78 mining industry is one of the most important contributors to the national economy. The
79 principal minerals found in the country include bauxite, chromite, copper, diamond, dolomite,
80 fluorite, graphite, gold, iron ore, limestone etc. Waste materials are generating in abundant
81 quantity during the mining of minerals and ores. Since they are no longer useful to the mine
82 owners, after the extraction of required material, tailings are being dumped in the surroundings
83 which occupy a large amount of land. If not properly maintained, they cause some major
84 hazards to the environment in the form of land degradation, fire hazards, water pollution and
85 air pollution due to the presence of toxic and combustible compositions¹⁻⁵. But maintaining the
86 stability of these dumps is also a major challenge for the mining industry. It has been reported
87 that the quantity of tailings generation went upto 90 – 98% for copper ores and 20 – 50% for
88 other minerals⁶⁻⁸. Thus, a huge quantities of tailings are being generated from different sources
89 such as copper ore, bauxite ore, iron ore and gold ore which are estimated to be 4 million
90 tonnes, 2.7 billion tonnes, 290 million tonnes, and 1 tonnes, respectively per annum^{9,10}.

91 Due to the lack of landfill space and its high cost, the usage of these mine tailings in some
92 application has become increasingly essential. However, the effective approach for the
93 application of these mine tailings is not being suitably developed^{1,2,5}. Thus, there is a potential
94 need to utilize these mine tailings in an effective manner in different fields like construction,
95 pavements, backfilling etc. Whereas on the other hand, river sand mining has been prohibited
96 in many provinces due to the adverse impacts causing ecological imbalances in the
97 environment. But natural river sand has been widely accepted as a conventional fine aggregate
98 in the production of cement mortars. As of now, fine crushed granite rocks (manufactured sand,
99 i.e., M-Sand) have largely replaced the river sand in construction industry. Even though M-
100 Sand showed satisfying performance as an alternative to river sand, it has certain disadvantages
101 in fulfilling the performance requirements of concrete such as grading requirements,
102 workability¹¹, surface finishing ability¹², compressive strength, density, yield etc. Moreover,
103 the problems associated with the river sand extraction are far more destructive than mining of
104 granite rocks which is being widely accepted as coarse aggregate in producing concrete^{13,14}.
105 Thus, the production of concrete and mortars using mine tailings is a best possible option where
106 these tailings can be employed sustainably as an alternative ingredient to river sand. The use
107 of these mine tailings as replacement to conventional materials can not only provide an
108 abundant and cheap source of raw materials for concrete or mortar, but also minimize pollutants
109 and improves the sustainability credentials.

110 The application of mine tailings as a replacement to river sand in cementitious system has been
111 investigated by the research fraternity and witnessed a satisfactory performance as the mine
112 tailings usually in a well-graded state with its particle size distribution being quite similar to
113 river sand. The previous researches have also revealed that the utilization of mine tailings is
114 economical, efficient, socially beneficial and improves environmental situation¹⁵⁻¹⁸. The

115 hardened properties of iron ore tailing (IOT) based cementitious composites were reported to
116 be increasing till 25%¹⁹, 25-40%²⁰, 35%²¹, 50%¹⁸ and even for 100%²² replacement of river
117 sand. Studies have been also reported that density of cementitious composites increases with
118 the usage of mine tailings in the presence of heavy metals^{15,18,20,23,24}. However, the workability
119 and durability characteristics were reported to be reducing with the usage of IOT^{18-20,22}.

120 The workability characteristics of cementitious composites were found to be enhanced by the
121 usage of copper ore tailing (COT) due to its particle morphology¹⁵. Previous studies have also
122 witnessed a slight improvement in the hardened properties of COT based cementitious
123 composites till 10-20%^{23,24}, 60%²⁵ and even till 100%²⁶ replacement of river sand. In contrast
124 to this, there is a study claiming the reduced hardened properties with the usage of COT¹⁵. In
125 addition to these, some studies have also confirmed the reducing durability characteristics by
126 the usage of COT^{23,24}. These observed hindrances were primarily attributed to inferior physical
127 properties and grading requirements of mine tailings in comparison with the conventional river
128 sand¹⁵.

129 However, there observed a great deal of variability in the reported values of engineering
130 properties in the case of cementitious composites produced by using mine tailings. Each of
131 them showed different impact on the mechanical properties and long-term durability of the
132 resulting cement mortar. The influence of using mine tailings followed by the mechanism
133 which alters the engineering properties of cement mortar were not thoroughly investigated in
134 the available literature. In addition, the characterization studies on these mine tailing based
135 cement mortars were found to be very limited in the literature and is need of the hour to
136 understand as it is very much essential in assessing the long-term behavior and to bridge the
137 future research. Hence, the present study aims to comprehensively investigate the influence of
138 using mine tailings as a substitute material to river sand through experimental investigations
139 and characterization techniques. Thus, an attempt has been made to discuss the physical and
140 chemical mechanism that occurs in the cementitious system upon the usage of mine tailings
141 that prerequisite the clear understanding of material, engineering, microstructure and durability
142 properties of cement mortar. Consequently, the present study contributes in developing a
143 sustainable solution for the problems associated with the waste management of copper and iron
144 ore tailings by effectively using them as an alternative to conventional fine aggregates in
145 practice, a way towards achieving sustainability.

146 **2. Materials and experimental methodology**

147 **2.1 Materials used and their properties**

148 The binder used for the production of cement mortar was 53 grade ordinary portland cement
149 (OPC 53G) conforming to ASTM C150/C150M-22²⁷. Locally available normal potable water
150 conforming to IS 10500:2012²⁸ was used for mixing and curing of mortar specimens. Natural
151 river sand falling under zone-II of IS: 383-2016²⁹ was utilized as fine aggregate and
152 polycarboxylate ether based superplasticizer conforming to IS 2645:2003³⁰ and IS 9103:1999
153³¹ was employed to achieve the required flow characteristics in the produced cement mortar.
154 The study mainly highlights the effective utilization of two varieties of mine tailings such as
155 iron ore tailings (IOT) and copper ore tailings (COT) as possible alternatives to river sand.

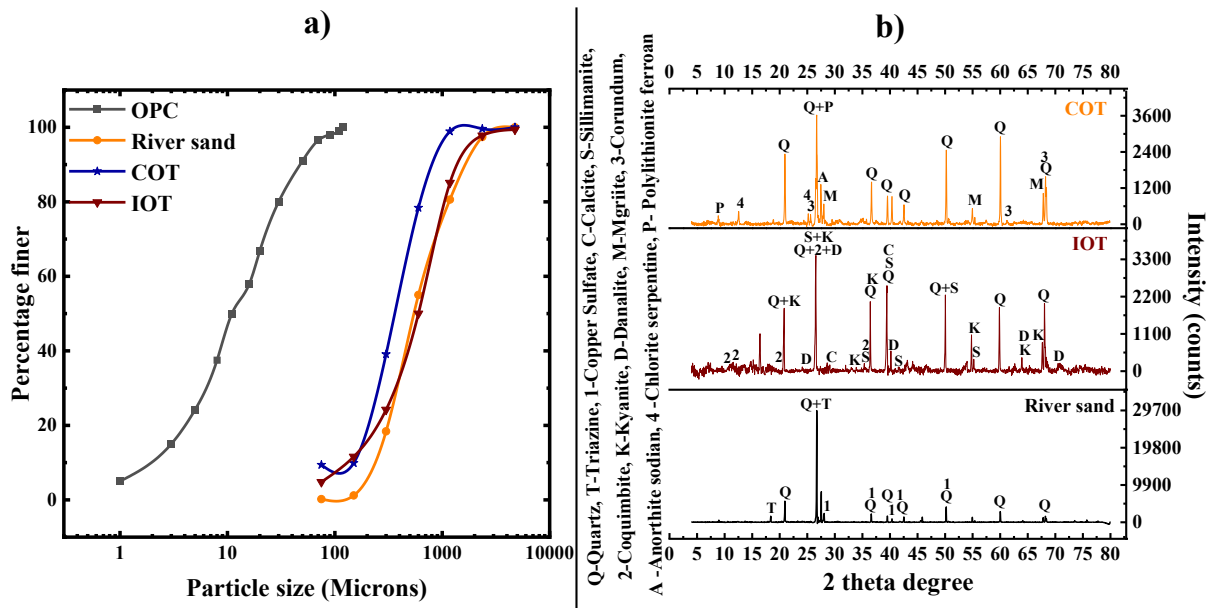
156 Various tests were conducted as per the standards to extract the properties of materials used
 157 and the observed results are tabulated in Table 1. Fig. 1a) presents their particle size distribution
 158 (PSD). The PSD data shows a significant volume of finer fractions (less than 150 microns)
 159 were present in mine tailings (11.6% in IOT and 9.9% in COT). The finer fractions of size
 160 below 75 microns were found to be 9.4% in COT and 4.8% in IOT. Fig. 1b) presents the XRD
 161 patterns of fine aggregates used which highlights their mineralogical compositions and particle
 162 geometry. The XRD patterns of river sand show highly crystalline geometry with majorly
 163 quartz and traces of copper sulphate and triazene in it. However, the tailing particles showed
 164 comparatively more amorphous structure than river sand due to the presence of reactive
 165 minerals in them. The lower intensity range of XRD peaks in tailing particles compared to that
 166 of river sand confirms the lower crystallinity of mine tailings. The analysis shows that the IOT
 167 has aluminosilicate minerals (Al_2SiO_5) such as sillimanite and Kyanite. The iron-based
 168 minerals such as coquimbite and Danalite were also detected through XRD analysis. Similarly,
 169 the COT shows aluminosilicate minerals such as polyolithionite, Anthorite ($CaAl_2Si_2O_8$),
 170 chlorite and traces of copper based mineral Mgriite. Similar kind of structural geometry was
 171 obtained in the past researches conducted on IOT^{19,22} and COT particles³²⁻³⁴.

172 The morphology of the tailing particles was studied by conducting SEM under secondary
 173 electron mode. Fig. 2 shows the scanning electron micrographs of mine tailings at different
 174 magnifications. Table 2 shows the chemical compositions of these mine tailings extracted from
 175 X-ray fluorescence spectroscopy (XRF) technique.

176 **Table 1: Properties of various materials used in the study**

Properties		Materials				
		OPC	RS*	IOT	COT	SP*
Specific gravity		3.11	2.59	3.32	2.39	1.08 ± 0.02
Fineness (m ² /kg)		300	-	-	-	-
Bulk density (kg/m ³)	Loose	-	1500	1733	1372	-
	Compacted	-	1598	1925	1575	-
% of Voids	Loose	-	43.07	47.71	42.61	-
	Compacted	-	39.34	41.93	34.10	-
Fineness Modulus		-	2.47	2.32	1.74	-
Water absorption (%)		-	0.98	0.66	0.79	-

177 *- RS: River sand, *SP- Superplasticizer

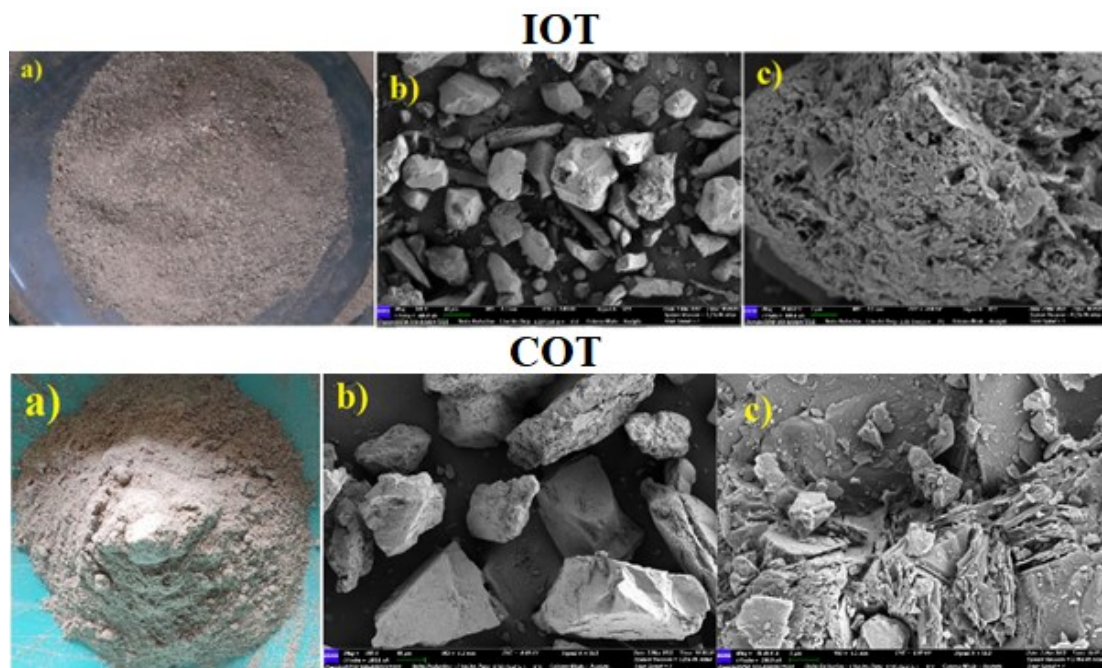


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Fig. 1. a) Particle size distribution of various materials used, b) Mineralogical composition of fine aggregates used



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Fig. 2. a) Mine tailing, b) scanning electron micrograph at 500X, c) scanning electron micrograph at 10,000X

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Table 2. Chemical compositions of IOT and COT

Component	(% mass)							
	MgO	Al ₂ O ₃	SiO ₂	P ₂ O ₅	SO ₃	K ₂ O	CaO	MnO
IOT	1.12	4.54	39.50	0.09	0.09	0.21	0.70	0.31
COT	2.03	8.68	65.26	0.27	4.33	4.12	3.68	0.10
Component	(% mass)							
	Fe ₂ O ₃	Na ₂ O	Cl	TiO ₂	CuO	SrO	ZrO ₂	

IOT	53.43	-	-	-	-	-	-
COT	9.65	0.69	0.01	0.64	0.39	0.06	0.07

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186 2.2. Experimental methodology

187 2.2.1 Details of cement mortar mixes developed

188 The study was fundamentally carried out with three varieties of cement mortar mixes. Cement
189 mortar produced without any replacement of river sand was considered as reference mix i.e.,
190 control mix (CM). The other two varieties of mortar mixes involve the replacement of river
191 sand with mine tailings. The mortar mixes nominated with alphabet “I” belongs to IOT based
192 mortar and with alphabet “C” belongs to COT based mortar. The subsequent numbers represent
193 the replacement level by volume of river sand with mine tailings. All the mortar mixes
194 developed in the study were prepared with water to cement ratio (w/c) of 0.45 and volume of
195 binder to fine aggregate ratio of 1:3. The necessary moisture and water absorption corrections
196 were incorporated while preparing the mortar mixes. To ensure better workability of produced
197 mortar mixes, a minimum flow value of 150 mm was maintained throughout the study using
198 superplasticizer. The detailed mix proportion along with the basic properties of various mortar
199 mixes produced in the present study are being tabulated in Table 3.

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Table 3: Details of the mortars produced

Mix name	Weight (kg/m ³)				SP (%)	Dry density (kg/m ³)		Setting time (mins)	
	OPC	RS	IOT/COT	Water		28 days	120 days	IST*	FST*
CM	777.50	1943.25	-	349.88	0.00	2247.82	2273.40	333	448
IOT based mortar									
I10	777.50	1748.93	248.63	349.88	0.09	2259.19	2311.29	288	445
I20	777.50	1554.60	497.25	349.88	0.10	2324.55	2385.17	283	439
I30	777.50	1360.30	745.88	349.88	0.10	2344.44	2387.07	269	423
I40	777.50	1165.95	994.50	349.88	0.25	2356.00	2389.34	227	412
I50	777.50	971.63	1243.13	349.88	0.40	2389.03	2415.49	219	409
COT based mortar									
C10	777.50	1748.93	179.25	349.88	0.00	2279.08	2311.29	304	436
C20	777.50	1554.60	358.50	349.88	0.10	2200.00	2239.30	302	426
C30	777.50	1360.30	537.75	349.88	0.10	2185.00	2225.09	291	420
C40	777.50	1165.95	717.00	349.88	0.10	2119.95	2224.52	275	404
C50	777.50	971.63	896.25	349.88	0.45	2057.43	2207.47	260	399

201 *bwob-by weight of binder; *IST-initial setting time; *FST-Final setting time

202 2.2.2 Production of cement mortar samples

203 Cement mortar specimens of size 70.6 mm × 70.6 mm × 70.6 mm were cast and tested in the
204 present investigation. Mixing of mortar was done in an automatic mortar mixer designed as per
205 the guidelines mentioned in EN 196-1³⁵. The cast specimens were further allowed to set under
206 humid environment (i.e., 27 ± 2°C and 95% RH). Further these specimens were demoulded

207 after 24 hours and immersed in normal potable water for curing maintained at $27 \pm 2^\circ\text{C}$ till the
 208 date of testing. After the curing period of 3, 7, 28, 56 and 120 days, samples were taken out,
 209 dried and tested to check their performance in hardened state.

210 2.2.3 Details of tests performed

211 2.2.3.1 Fresh and hardened properties

212 The mortar mixes were assessed in fresh state to check the workability and setting time. To
 213 investigate the hardened properties of produced mortars, tests were carried out using both
 214 destructive and non-destructive methods. These tests were conducted at the age of 3, 7, 28, 56
 215 and 120 days of curing for all the mortar mixes. The average of three values obtained from
 216 three specimens was taken into consideration and the same was reported. Table 4 shows the
 217 details of the tests performed to assess the fresh and hardened properties.

218 **Table 4. Details of the tests conducted to investigate the fresh and hardened properties**

Type	Test	Reference	Remarks
Fresh properties	Mini flow table test	EN 1015-3 ³⁶	-Workability/flowability measurement -To fix the superplasticizer dosage required for maintaining a minimum flow diameter of 150 mm
	Setting time	ASTM C403/C403 M-08 ³⁷	-Initial and final setting time measurement using Penetration resistance measurements
Hardened properties	Compressive Strength	IS 4031-Part 6:1988 ³⁸	-Using compressive testing machine at a loading rate of 35 MPa/min
	Dry density	BS EN 1015-10:1999 ³⁹	-Oven dried weight by volume method using weighing balance
	Permeable porosity	40,41	-Vacuum water absorption technique using vacuum desiccators
	Water absorption	ASTM C948 – 81 ⁴²	-Using change in weight at saturated surface dried and oven dried condition

219

220 2.3.3.2 Microstructural analysis and characterization techniques used

221 The performance of cement mortars was primarily influenced by their microstructure. Thus,
 222 understanding the microstructure of cement mortars and hydration products formed are very
 223 critical. The chunks were collected from the core of mortar samples after 28 and 120 days of
 224 curing and kept immersed in isopropyl alcohol for 24 hours to discontinue the hydration
 225 mechanism beyond that specific curing age^{43,44}. Further the chunks were dried in oven at a
 226 temperature range of 40-60 °C till the weight gets stabilized. These oven dried samples were
 227 then stored in moist free desiccators maintained using silica gel pellets and vacuum pump.
 228 These samples were examined using various characterization techniques to investigate the
 229 microstructure development in various types of cement mortars.

230 a) Scanning electron microscopy and energy dispersive X-Ray spectroscopy (SEM-EDS)

231 In order to investigate the developed microstructure and to carry out the morphological
232 observations in optimized mortars (best performing), scanning electron microscopy (SEM)
233 technique was used in secondary electron mode. Further, with the aid of energy dispersive X-
234 Ray spectroscopy (EDS), the elemental composition of various types of cement mortars were
235 analyzed. All the preserved samples were gold sputtered before taking them for Microstructural
236 examination.

237 b) X-Ray diffraction (XRD) analysis

238 The mineralogical compositions of produced mortars were studied with the help of X-Ray
239 diffraction (XRD) studies. The preserved samples were ground and sieved in 75 microns IS
240 sieve and the collected powdered fine particles were taken for examination. The studies were
241 carried out in Jeol-JPX 8P and Malvern Panalytical make X-ray diffractometers equipped with
242 Cu Ka radiation (40 kV/40 mA) at a scanning rate of 2°/min. The samples were examined with
243 the 2θ angle ranging from 4° to 80° and the received XRD patterns were further analyzed with
244 the aid of software X'Pert High Score Plus.

245 c) Thermogravimetric analysis (TGA)

246 Thermogravimetric analysis was performed using RIGAKU TG-DTA 8112 analyzer to
247 quantify the hydration products formed in different types of mortars at 28 days and 120 days
248 of curing age. The hardened paste from the preserved mortar samples were ground in to
249 powder, sieved through 75 μm sieve. The finer powder collected after sieving was
250 characterized in a nitrogen purge environment at temperatures ranging from room temperature
251 to 875°C. The heating rate was maintained at 20°C/min and purge rate was maintained at 20
252 ml/min. When the hydrated mortar samples were subjected to elevated temperatures,
253 Thermogravimetric mass loss takes place at various different temperature boundaries due to
254 the evaporation of free water, decomposition of hydration products by dehydration, de-
255 hydroxylation and de-carbonation^{43,45,46}. These temperature boundaries were determined based
256 on the endothermic peaks formed in derivative thermogravimetric curve (DTG) due to
257 decomposition (weight loss) of compounds. The endothermic peak usually occurs in the
258 temperature range of 50-120°C, 120-150°C, 110-300°C and 230-380°C corresponds to the
259 dehydration of water molecules from the ettringite, gypsum, CSH and friedel's salt,
260 respectively^{43,47-52}. Thereafter the weight loss that occurs between 400-500°C is associated to
261 the decomposition of calcium hydroxide (CH)^{43,48}. Similarly, the temperature boundary
262 associated to the decomposition of calcium carbonate appears in the temperature boundary of
263 600-800°C⁴³.

264 d) Fourier transform infrared spectroscopy (FTIR)

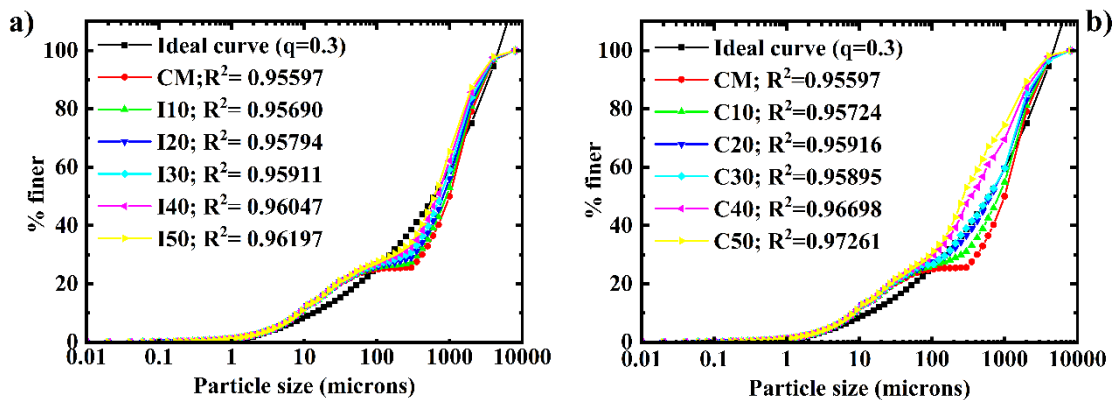
265 FTIR studies were performed with the aid of Bruker Alpha II FTIR equipment and OPUS
266 software. The sample preparation is similar to that followed for XRD and TGA analysis. FTIR
267 studies were performed to obtain the transmittance spectra in the wavelength ranges from 600
268 cm⁻¹ to 4000 cm⁻¹ under ATR mode. The formation of different phases in cementitious system
269 such as portlandite (Ca (OH)₂), C-S-H, CASH, sulfates and carbonate phases (such as ettringite,
270 hemcarbonate, monosulfoaluminate, sulfoaluminate, calcium carbonate) etc. can be recorded

271 based on the peak intensity and shift over the age of curing and also compared among the
272 different types of mortars produced.

273 3. Analysis of results and discussions

274 3.1 Particle packing of the produced cement mortar mixtures

275 The particle packing of eleven number of mortar mixes designed in the present study were
276 analyzed using modified Andreasen and Andersen packing model⁵³. In order to carry out the
277 analysis, “EMMA” software was employed which adopts the principle of modified Andreasen
278 and Andersen packing model. Fig. 3 shows the particle packing curves of mine tailing based
279 mortars. A distribution coefficient (q) of 0.30 appropriate for achieving medium workability
280 was selected for the analysis⁴³. From the obtained curves, it is can be clearly observable that
281 the addition of mine tailings improves the particle packing of mortars till a specific replacement
282 level by improving the extent of fitting to the target curve fixed by the modified Andreasen and
283 Andersen packing model for highest particle packing density⁵⁴. Among the various mortar
284 mixes designed, I20 and C30 mixes showed best fitting to the target curve.



285

286 **Fig. 3. Particle packing curves of a) IOT based mortars and b) COT based mortars**
287 **determined using modified Andreasen and Andersen particle packing model**

288 3.2 Fresh properties

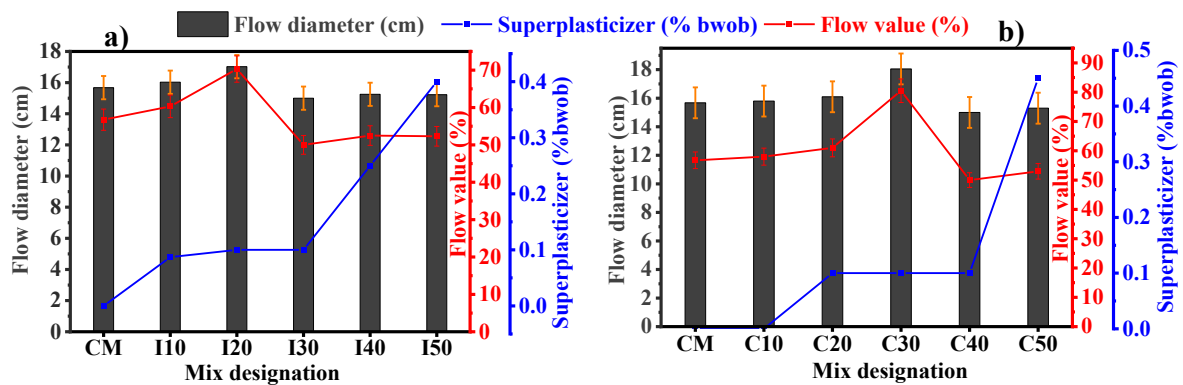
289 3.2.1 Workability

290 Workability of different types of cement mortar mixes produced using varying dosages of IOT
291 and COT as a replacement to river sand were checked using mini flow table test. To ensure the
292 desired workability in the produced mortar mixes, a minimum flow diameter of 150 mm was
293 fixed and achieved with the aid of superplasticizer. The results reported in the Fig. 4 a) (IOT
294 based mortar) and Fig. 4 b) (COT based mortar) consist of flow diameter, flow value and their
295 corresponding dosage of superplasticizer consumption.

296 From the obtained results, it can be noticeable that the addition of IOT into the mortar mix as
297 a replacement to river sand reduces the flow characteristics. This behavior can be clearly
298 observed through the dosage of superplasticizer consumption with the percentage usage of IOT.
299 The control mortar (CM) produced entirely with river sand as fine aggregate was observed to
300 achieve required minimum flow diameter of 150 mm without consuming any superplasticizer.
301 With the percentage usage of IOT as replacement to river sand, the flow characteristics were
302 observed to become poorer due to the intrinsic properties of IOT. The rough textured IOT
303 particles increase the friction between the particles and the sharp edges of IOT particles block

304 the movement of particles over one another. Also, the fine nature of IOT particles as compared
 305 to river sand leads to filling the void spaces and make the mortar more uniform and cohesive
 306 till I20. Further there observed a drastic fall in the flow values (i.e., from I30) which can be
 307 attributed to the poor gradational characteristics due to the occurrence of excessive amount of
 308 fines which demands higher water content for surface wetting resulting in the formation of
 309 stiffer mortar mixes. These observations are in line with the inference made in some past
 310 researches^{19,20}. The dry and harsh mortar mixes thus produced leads to increased
 311 superplasticizer consumption to a greater extent and thereby mortar mixes showed reduced
 312 workability.

313 Similarly in the COT based mortar, a sudden drop in the flow values were observed beyond
 314 30% replacement of river sand with COT due to the increased amount of fines in the mortar
 315 mix. This can be clearly observed through the flow values and the superplasticizer dosages
 316 consumed among the mixes C30, C40 and C50. Even though the superplasticizer dosage was
 317 kept constant between the mixes C30 and C40, there observed a sudden declination in the flow
 318 values and further from C40 to C50 mix, there observed a sudden increment in the
 319 superplasticizer dosage requirement. However, the flow characteristics were found to be
 320 improved right from the CM and till C30. This can be clearly noticeable from the flow values
 321 and their corresponding superplasticizer dosage from CM to C10 and C20 to C30 mix. This
 322 behavior can be attributed to the smooth glossy textured and angular shaped COT particles
 323 reduce the friction and helps in movement among the particles with great ease^{25,26}. Also, the
 324 fine particles of COT initially tend to fill the voids at lower replacement levels (till 30%) and
 325 make the mixes uniform and cohesive that helped in improving the flow characteristics¹⁵.



326

327 **Fig. 4. a) Flow characteristics of a) IOT based mortar; b) COT based mortar**

328 **3.3 Hardened properties**

329 **3.3.1 Compressive strength**

330 Compressive strength development in control (CM), mine tailing based mortars were measured
 331 at the curing age of 3, 7, 28, 56 and 120 days and the results are plotted as histogram. Fig. 5 a)
 332 and Fig. 5 b) presents the compressive strength of IOT based and COT based mortars,
 333 respectively.

334 The results obtained from both IOT and COT based mortars exhibited a decreasing trend with
 335 the content of mine tailings in the initial days of curing (i.e., 3 days and 7 days). Even though
 336 mine tailings accelerated the hydration reaction leading to early setting of mortar (Table 3),

337 they exhibited slightly lower compressive strength at early ages of curing till 7 days. This
338 reduction in the compressive strength with the increasing amount of mine tailings is due to the
339 slowing down of hydration reaction by the development of low permeability layer of heavy
340 metals around the non-hydrated cement grains^{55,56}. The presence of Cu ions also slightly delays
341 the hydration process due to the precipitation of oxides and hydroxides⁵⁶⁻⁶⁰.

342 However there observed a significant improvement in the compressive strength development
343 of mine tailing based mortars with the progress in curing age (beyond 7 days of curing). This
344 could be attributed to the acceleration in the hydration activity of cement particles under the
345 influence reactive minerals present in tailings. The presence of iron compounds at higher
346 concentrations in the mortar accelerates the hydration process^{19,57,58,61,62}. The additional
347 calcium content added from the finer fractions of mine tailings also contributes in accelerating
348 the hydration process^{57,58}. The IOT based mortars showed an increasing trend of compressive
349 strength till a sand replacement level of 20% and COT based mortars till a sand replacement
350 level of 30%. The improved particle packing and thereby improvised particle packing density
351 of the mortar mixes with the addition of IOT till 20% and COT till 30% helped the ingredients
352 of mortar to undergo primary and possible secondary hydration reactions more effectually^{19,63}.
353 Since the mine tailings are rich in siliceous and aluminous compounds, the finer fractions of
354 mine tailings facilitate in undergoing supplementary hydration reactions (secondary hydration
355 reaction/ pozzolanic reaction) after 7 days of curing resulting in the formation of additional
356 hydration products¹⁹.

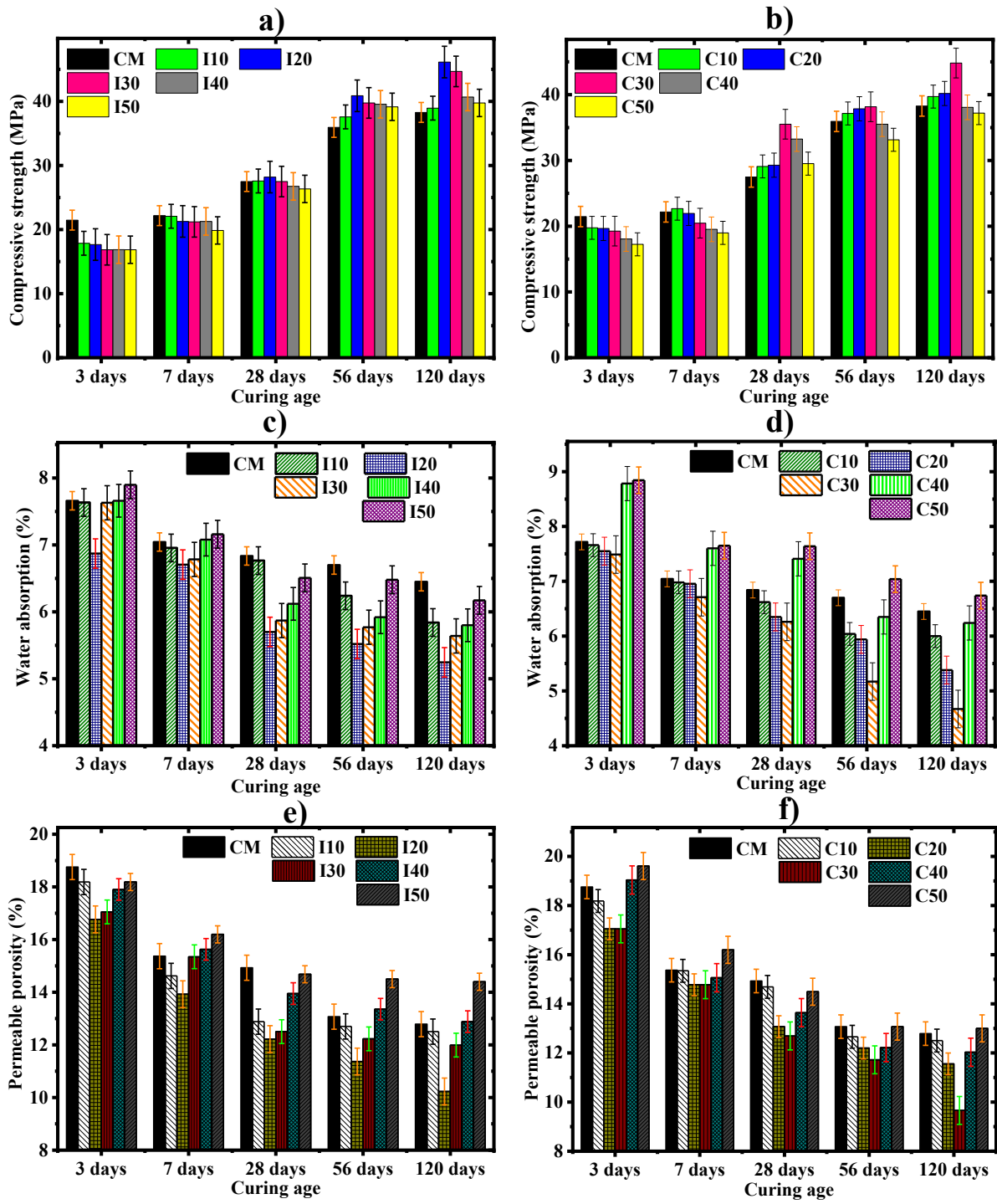
357 The dry density results presented in Table 3 also evident the occurrence of secondary hydration
358 reaction. The difference between the dry densities of mine tailing based mortars with the
359 percentage replacement of sand was found to be reducing with the progress in curing age which
360 is attributed to the formation of additional hydration products by the finer fractions of tailings
361 through secondary hydration reactions. These actions helped in refining the pore structure of
362 mortars in an effective manner leading to the formation of dense microstructure that enhances
363 the hardened mechanical properties such as compressive strength. It is also important to notice
364 from the results that the compressive strength of the mortars reduces beyond the optimized
365 dosage of mine tailings which is due to the effect from poorer gradational characteristics and
366 excessive silica-alumina content that restrains the degree of hydration reactions. Similar trend
367 of compressive strength of mortar by the usage of IOT & COT in relation to control mortar
368 was observed in the past literatures^{18-20,23-25}. The percentage strength gain of I20 mortar was
369 recorded to be 2.55%, 13.75% and 20.53% at the curing age of 28, 56 and 120 days
370 respectively. Similarly, C30 mortar exhibits 29.17%, 6.20% and 17.08% at 28, 56 and 120 days
371 of curing age respectively.

372 3.3.2 Water absorption and Permeable porosity

373 The water absorption and permeable porosity of control and mine tailing based mortars were
374 assessed at the end of 3, 7, 28, 56 and 120 days of curing. Fig. 5 c) and Fig. 5 d) presents the
375 water absorption of IOT based and COT based mortars respectively. Similarly, Fig. 5 e) and
376 Fig. 5 f) represents the permeable porosity of IOT based and COT based mortars respectively.
377 The results obtained shows that both water absorption and permeability porosity follow a quite
378 similar trend with the percentage usage of mine tailings at all the curing ages. This could be

379 due to the fact that the absorbed water predominantly accumulates in the permeable void spaces
380 or pores present in the mortar mixes and these pore volumes are usually regarded as porosity.
381 It is also important to notice that both water absorption and permeable porosity reduces with
382 the progress in curing age which is due to the formation and accumulation of hydration products
383 that refines the pore structure in mortar.

384 The utilization of mine tailings in the mortars as a replacement to river sand showed positive
385 results in the case of water absorption and permeable porosity. The water absorption and
386 permeable porosity were found to be reducing with the increasing proportion of mine tailings
387 till certain replacement level and thereafter found to be increasing. This behavior can be
388 ascribed to the improved particle packing and improved affinity to undergo secondary
389 hydration reaction by the finer fractions of mine tailings with the increasing level of sand
390 replacement^{19,63}. Similar trend of reducing water absorption with the usage of IOT was noticed
391 by previous studies¹⁹. The mortar mixes “I20” and “C30” recorded lowest water absorption
392 and permeable porosity values. In addition to these, the lower water absorbing capacity of IOT
393 and COT in comparison with the river sand also contributes in reducing the water absorption
394 and permeable porosity of mortar. However, exceeding the dosage of mine tailings beyond
395 optimum percentage (i.e., 20% of IOT and 30% of COT) causes an increase in the water
396 absorption and permeable porosity due to poor gradational characteristics and lack in water
397 availability for undergoing hydration reactions caused by the increased amount of fines in the
398 mortar mix. The reduced alkalinity in the mortar by the high-volume replacement of sand by
399 mine tailings might also be the cause for the increased water absorption and permeable
400 porosity.



401

402 **Fig. 5. a) Compressive strength, c) Water absorption, e) permeable porosity of IOT**
 403 **based mortars, and b) Compressive strength, d) Water absorption, f) Permeable**
 404 **porosity of COT based mortars**

405 3.4 Microstructure and characterization studies

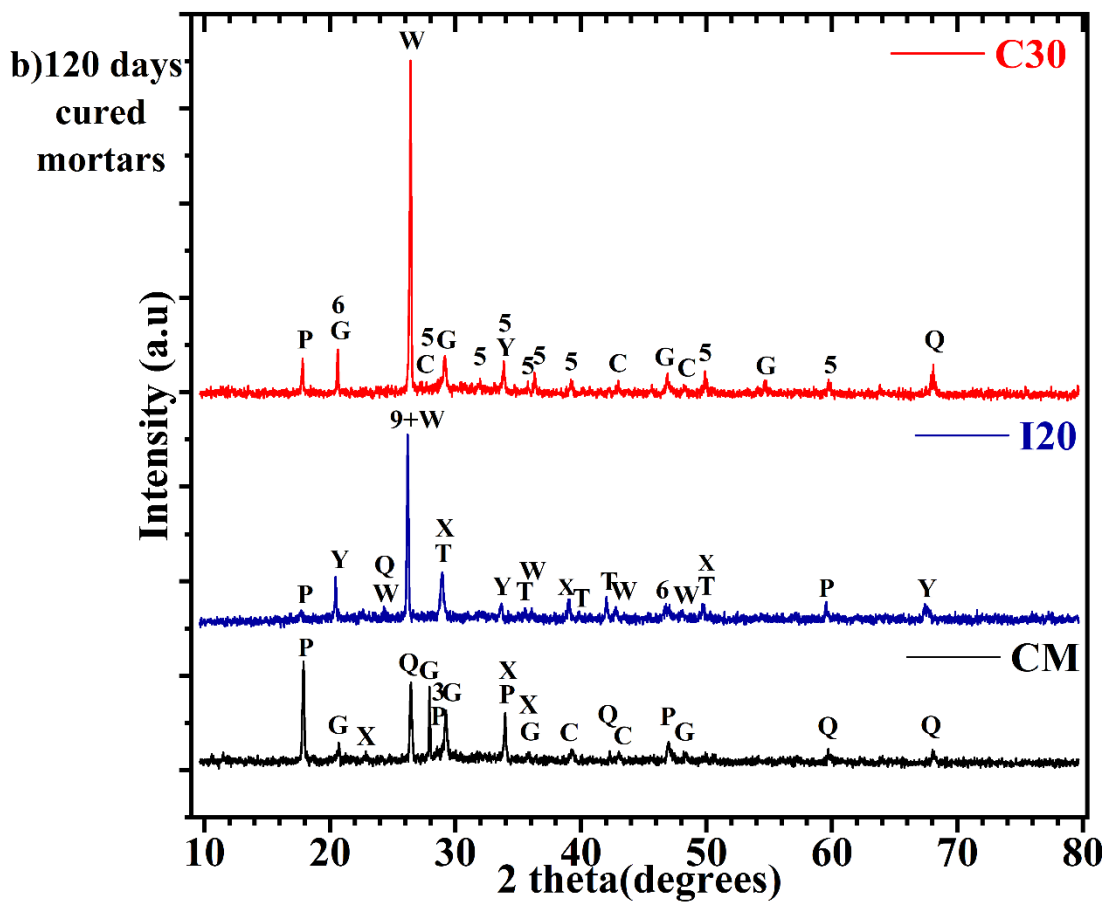
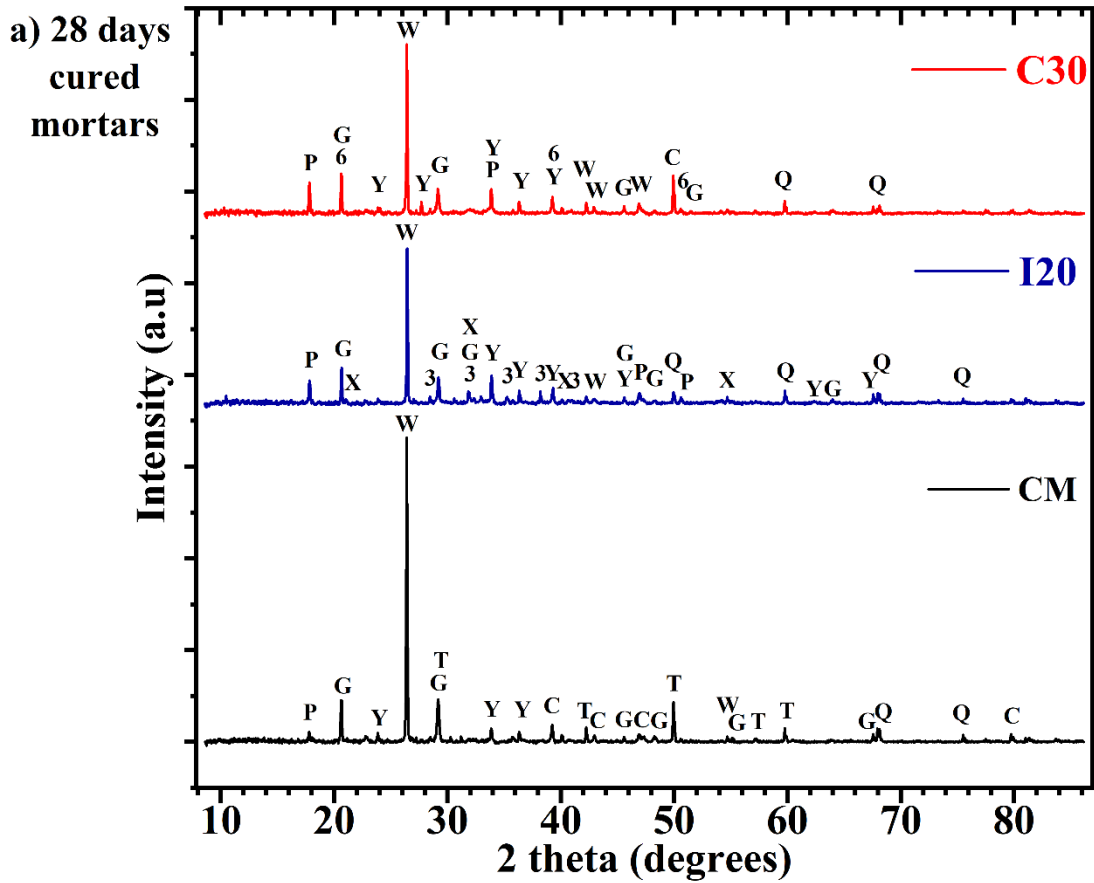
406 3.4.1 Mineralogical characterization

407 The influence of different dosages of mine tailings on the mineralogical compounds was
 408 studied at the curing age of 28 days and 120 days using the XRD patterns. The XRD analysis

409 was carried out for the control mortar, and till one replacement level beyond the optimum mine
410 tailing-based mortar (i.e., till I30 and C40). The details of the mineralogical compounds such
411 as chemical name, empirical formula, peak assignment designations were tabulated in Table 5.

412 Fig. 6 a) presents the XRD patterns of 28 days cured and Fig. 6 b) presents XRD patterns of
413 120 days cured mortar samples of the optimized mixes (i.e., CM, I20 and C30). By observing
414 the XRD patterns of 28 days cured mortar samples, it can be noticeable that the crystalline
415 phases are more and prominent in mine tailing based mortars due to the presence of more CH
416 and other unhydrated particles in comparison with the control mortar. The occurrence of
417 portlandite (CH) peaks are more and prominent in mine tailing-based mortars which is due to
418 additional portlandite formed from the calcium particles present in the finer fraction of mine
419 tailings^{57,58}. The increased portlandite peaks is also due to accelerated hydration reaction
420 caused by improved particle packing intensity and higher iron oxide compounds^{57,58,61-63}. In
421 addition to these, there observed a preferential formation of additional calcium silicate hydrates
422 (CSH) and calcium silicate aluminum hydrate (CASH) phases due to the occurrence of
423 effective primary and secondary hydration reactions facilitated by the finer fraction of mine
424 tailings. The mineralogical compounds related to CSH and CASH phases such as Gismondine,
425 Chabazite, Xonotlite and Hillebrandite were found to be additionally formed.

426 However, by observing the XRD patterns of 120 days cured mortar samples, it can be
427 noticeable that the XRD patterns of mine tailing based mortars appeared more amorphous than
428 28 days ones and also in comparison with the control mortar. This is attributed to the effective
429 conversion of crystalline CH to amorphous gel like structure formed by CSH and CASH phases
430 through primary and secondary hydration reactions. The produced CH in control mortar
431 remained mostly unutilized after 28 days of curing. Whereas, the portlandite produced in the
432 mine tailing-based mortars were appeared to get consumed for secondary hydration reactions.
433 The formation of additional hydration phases related to CSH and CASH are evident for the
434 consumption of portlandite in mine tailing-based mortars. The mineralogical compounds
435 related to CSH and CASH phases such as Gismondine, Chabazite, Wairakite, Hillebrandite and
436 Tobermorite were found to be additionally formed which helped in refining the pore structure
437 of mine tailing based cement mortar. In addition, the occurrence of various phases of
438 carbonates, gypsum and ettringites are also the cause for more crystalline XRD patterns in
439 control mortar.



441 **Fig. 6. XRD pattern of optimized mortars at the curing age of a) 28 days and b) 120**
442 **days**

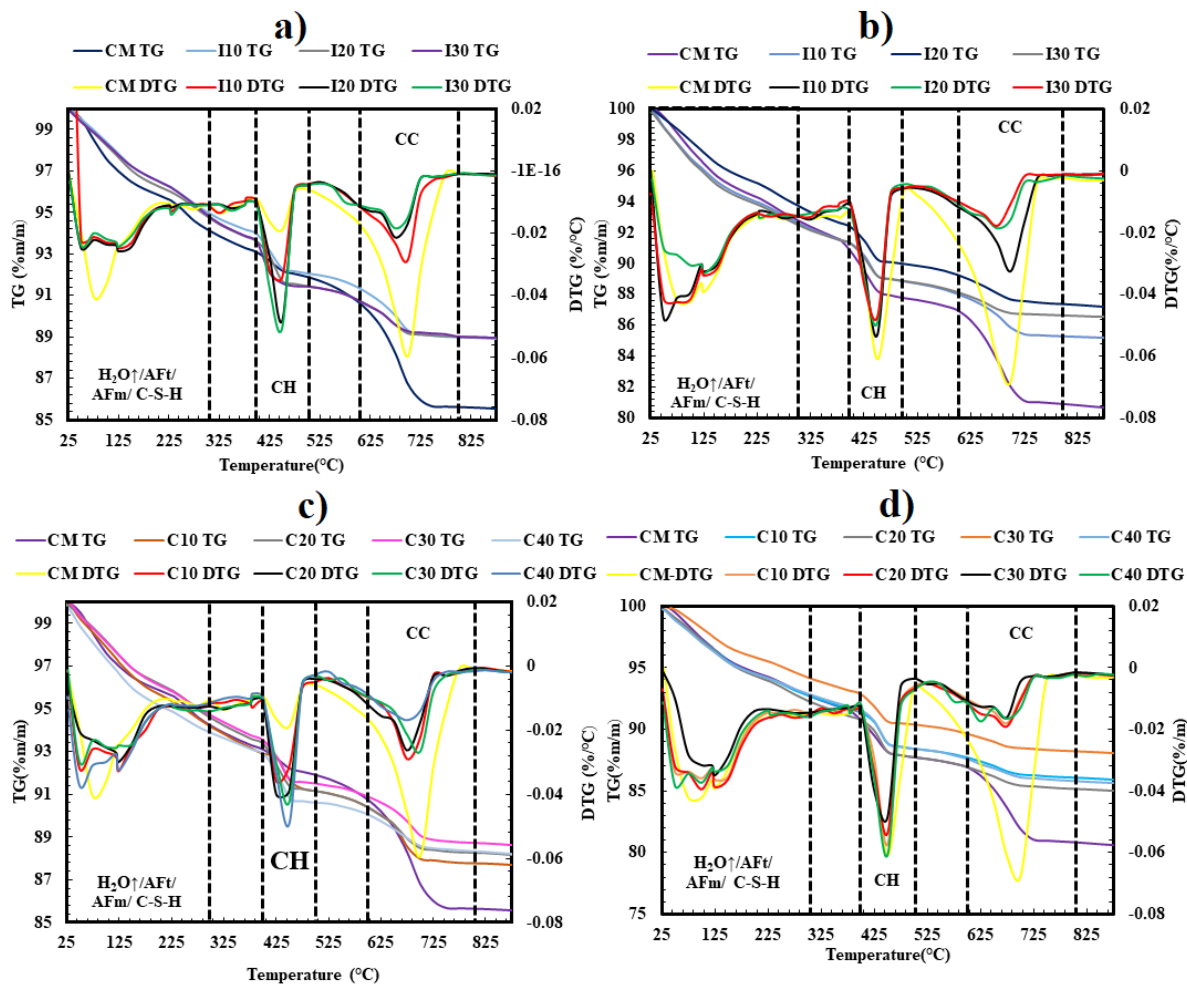
443 Overall, by observing the XRD patterns, there can be seen that more mineralogical compounds
444 related to calcium silicates and calcium aluminum silicates were formed with lesser carbonates,
445 gypsum and ettringite phases by the utilization of mine tailings in cement mortar. This
446 development in mine tailing mortars helped to perform superior than the control mortar. This
447 enhanced performance can also be witnessed through the measured engineering properties such
448 as compressive strength, water absorption and permeable porosity.

Table 5. Details of the mineralogical compounds and their peak assignment

Compound details	Peak assignment of COT based mortar										Peak assignment of IOT based mortar					
	28 days curing					120 days curing					28 days curing			120 days curing		
	CM	C10	C20	C30	C40	CM	C10	C20	C30	C40	I10	I20	I30	I10	I20	I30
Quartz (Silicon Oxide) (SiO ₂)	Q	Q	Q	Q	Q	Q	Q	Q	Q	Q	Q	Q	Q	Q	Q	Q
Portlandite (Calcium Hydroxide) (Ca (OH) ₂)	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P	P
Hillebrandite (Calcium Silicate Hydroxide) (Ca ₂ (SiO ₃) (OH) ₂)			2		2								2			2
Kamaishilite (Ca ₂ Al ₂ SiO ₆ (OH) ₂) (Calcium Aluminium Silicate Hydroxide)		3				3					3	3				
Tobermorite (Ca ₅ (OH) ₂ Si ₆ O ₁₆ ·4H ₂ O) (Calcium Silicate Hydroxide Hydrate)	T														T	
Xonotlite (Calcium Silicate Hydrate) (Ca ₆ Si ₆ O ₁₇ (OH) ₂)			X		X	X	X			X	X	X		X	X	
Calcium Silicate Hydrate (Ca ₂ SiO ₄ ·H ₂ O)			5				5	5	5							
Chabazite (Ca ₂ Al ₄ Si ₈ O ₂₄ ·12H ₂ O) (Calcium Aluminum Silicate Hydrate)				6	6				6						6	
Wairakite (CaAl ₂ Si ₄ O ₁₂ ·2H ₂ O) (Calcium Aluminum Silicate Hydrate)	W			W					W			W			W	
Gismondine (CaAl ₂ Si ₂ O ₈ ·4H ₂ O) (Calcium Aluminium Silicate Hydrate)		7	7				7	7		7	7		7	7		7
Calcite (Calcium Carbonate) (CaCO ₃)	C	C	C	C	C	C	C	C	C	C	C		C	C		C
Gypsum (Calcium Sulfate Hydrate) (CaSO ₄ ·2H ₂ O)	G			G	G	G	G	G	G		G	G	G		G	
Yeelimite (Calcium Aluminum Oxide Sulfate) (Ca ₃ Al ₆ O ₁₂ ·CaSO ₄)	Y			Y					Y	Y		Y		Y	Y	

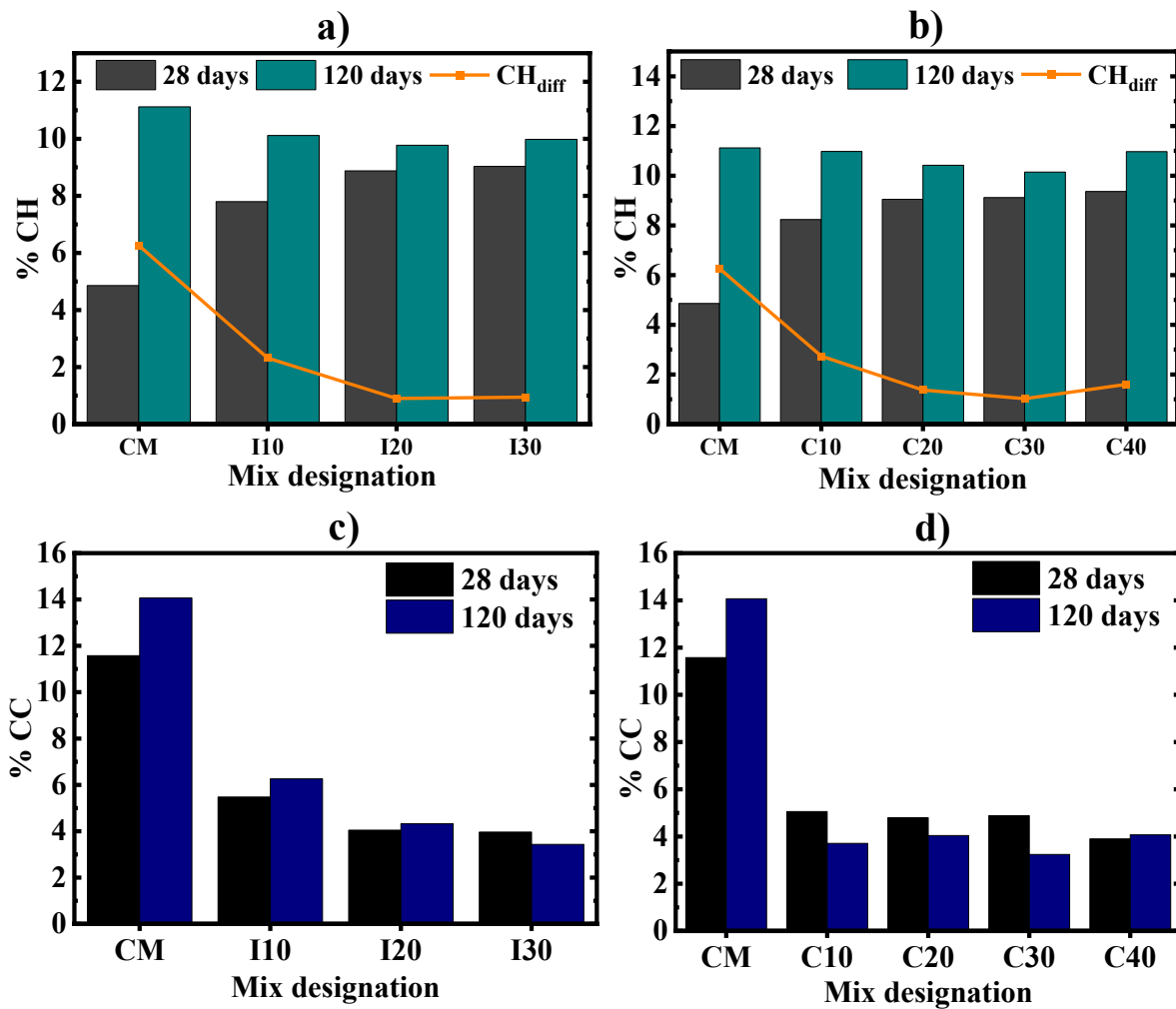
451 3.4.2 Thermogravimetric analysis (TGA) on IOT and COT based cement mortar

452 From the thermogravimetric analyzer, the weight loss of the mortar samples with the increasing
 453 temperature was recorded and for the analysis TG/DTG curves were plotted. Fig. 7 a) and Fig.
 454 7 b) presents the TG/DTG curves of 28 days and 120 days cured IOT based mortar samples,
 455 respectively. Similarly, Fig. 7 c) and Fig. 7 d) shows the TG/DTG curves of 28 days and 120
 456 days cured COT based mortar samples, respectively.



457
 458 **Fig. 7. TG and DTG curves of a) IOT based mortar after 28 days of curing, b) IOT**
 459 **based mortar after 120 days of curing, c) COT based mortar after 28 days of curing and**
 460 **d) COT based mortar after 120 days of curing**

461 In the present study, the hydration products such as calcium hydroxide (CH) and calcium
 462 carbonate (CC) were quantified on the basis of their decomposition by weight loss at their
 463 specific temperature boundaries. The various other hydration products such as gypsum,
 464 ettringite, CSH, CASH, friedels salts etc. were not computed in the present study due to the
 465 overlapping among the respective temperature boundaries^{43,49,50}. The quantified values of CH
 466 and CC are graphically represented in Fig. 8.



467

468 **Fig. 8. Phase compounds formed in IOT based mortar [(a) CH; (c) CC] and COT based**
 469 **mortar [(b) CH; (d) CC] after 28 days and 120 days of curing**

470 The TGA results of 28 days cured IOT and COT based mortar samples shows an increasing
 471 CH content with the increasing percentage of river sand replacement by mine tailings. This is
 472 attributed to the accelerated hydration reaction due to enhanced particle packing density and
 473 participation of elements of iron, i.e., Fe⁺ generated from the finer fractions of mine
 474 tailings^{19,57,58,61,62}. In addition to this, the additional CaO added from the mine tailings also
 475 marginally contributes to the generation of CH^{57,58}. Meanwhile, there will be some influence
 476 of heavy metals and copper ions too. The heavy metals present in mine tailings create a low
 477 permeability layer around the particles which delays the hydration reaction^{55,56}. The presence
 478 of Cu ions also slightly delays the hydration process due to the precipitation of oxides and
 479 hydroxides⁵⁶⁻⁶⁰. The delayed hydration reaction thus slows down the consumption of CH for
 480 further reactions which creates a higher CH reserve in the mine tailing-based mortars pore
 481 solution. The improved particle packing caused by the finer particle size distribution of mine
 482 tailings also leads to the extensive formation of CSH/CASH gels along with higher volume of
 483 portlandite^{57,58,61-63}.

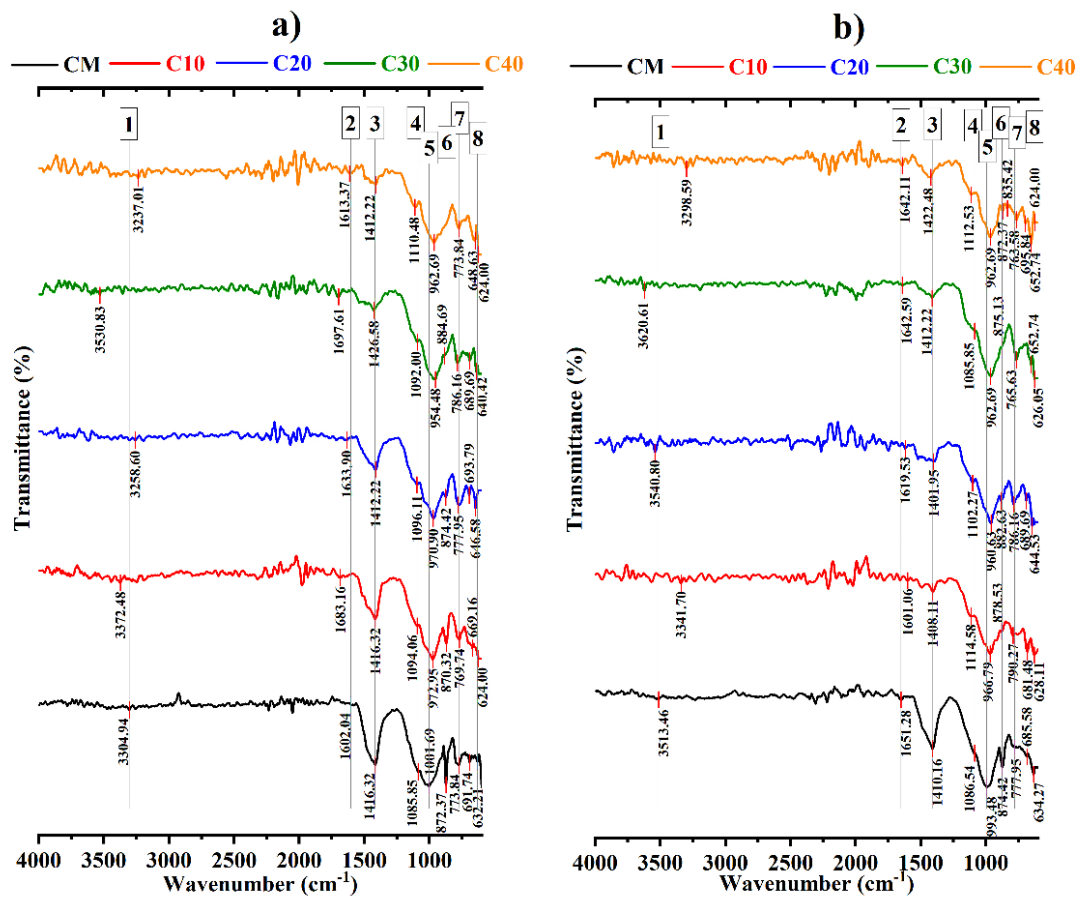
484 Apart from these, mine tailing based mortars recorded lower content of CC which can be seen
 485 in Fig. 8 where the control mortar showed highest amount of CC. It is also essential to notice

486 that there is a preferential formation of ettringite in the control mortar which can be seen by an
487 intensive endothermic peak in DTG curve. However, the mine tailing-based mortar showed
488 comparatively wider but less intense endothermic peak which is evident for the lesser formation
489 of gypsum and ettringite.

490 By observing the DTG curves and quantified CH values corresponding to 120 days cured
491 samples, it can be seen that the generation of CH continues with the curing age in all types of
492 mortars. Whereas the consumption of CH for the secondary hydration reactions was found to
493 be significantly more in mine tailing based mortars which can be seen from the reducing trend
494 of CH_{diff} (difference between the CH values at 28 and 120 days curing) shown in Fig. 8.
495 Conversely the endothermic peaks corresponding to CSH and CASH were found to be
496 broadened and intensified with the curing age which signifies the effective formation of CSH
497 and CASH gels in mine tailings based mortars. It is also important to note that the consumption
498 of CH is slightly more in the optimized mine tailing based mortar (i.e. I20 and C30) which is
499 due to the optimized mine tailing percentage favorable for undergoing reactions and superior
500 particle packing that might helped in undergoing the reactions efficiently. The quantified values
501 of CC by the end of 120 days of curing also shows the presence of CC in lower volume in mine
502 tailing based mortars. Thus, the additionally formed CSH/CASH gels along with the lower
503 content of CC, ettringite and gypsum ensures better performance in the case of mine tailing
504 based mortars.

505 3.4.3 Fourier transform infrared spectroscopy (FTIR)

506 Fig. 9 a) and Fig. 9 b) illustrates the FTIR spectra of IOT based cement mortar at the curing
507 age of 28 days and 120 days respectively. Similarly, Fig. 10 a) and Fig. 10 b) illustrates the
508 FTIR spectra of COT based cement mortar at the curing age of 28 days and 120 days
509 respectively. The corresponding details about the functional group assignment for various
510 wavenumbers are presented in Table 6.



514

515 **Fig. 10. FTIR spectra of control and COT based mortar samples at a) 28 days of curing**
 516 **and b) 120 days of curing**

517 It can be observed from the FTIR spectra's that the vibration bands and the corresponding
 518 functional groups assigned were found to be same in all the types of mortars (i.e., CM, IOT
 519 based and COT based mortar) both at 28 days and 120 days of curing. By observing the FTIR
 520 spectra's, it can be noticeable that the vibration band corresponding to Si-O-Si asymmetric
 521 stretching found in the wavenumber range of 1000 cm^{-1} is more prominent in control mortar
 522 and optimized mine tailing mortars (i.e., I20 and C30). These vibration bands were found to be
 523 more intense and broader than the rest of the mine tailing based mortars⁶⁴⁻⁶⁶. This could be
 524 attributed to the uninterrupted hydration reactions occurred in CM and effective hydration
 525 reactions taken place under the effect of superior particle packing along with the presence of
 526 iron oxides mine tailings, that facilitated in the generation of higher amount of CSH gels⁶³. The
 527 occurrence of secondary hydration reactions facilitated by the finer fractions of mine tailings
 528 also generates additional C-S-H phases.

529 For brevity, the wavenumbers corresponding to Si-O-Si/Al asymmetric stretching were color
 530 scaled from red to green in Table 6, where red represents the highest, green represents the
 531 lowest and yellow represents the wavenumber that exists between the highest and lowest. By
 532 observing the wavenumbers corresponding to Si-O-Si/Al asymmetric stretching band, there
 533 observed a change in color scale from red to green by the utilization of mine tailings which
 534 signifies the decrease in wavenumbers from 1001.68 cm^{-1} to 954.48 cm^{-1} in 28 days cured

535 mortars and from 993.48 cm^{-1} to 960.63 cm^{-1} in 120 days cured mortars. The decrease in
536 wavenumber of Si-O-Si/Al asymmetric stretching band signifies the change of microstructure
537 followed by the formation of more amorphous products with stronger bonds due to the
538 additional formations of silicates and aluminosilicate hydrates⁶⁷⁻⁷⁰. It is also important to note
539 that the I10 and I30 mortars showed additional peaks at wavenumbers 993.48 cm^{-1} and 997.58
540 cm^{-1} respectively which is evident for the formation of additional C-S-H phases by the intrusion
541 of IOT in mortar.

542 The vibration band associated to Si-O-Si/Al symmetric stretching were found to be more
543 prominent in mine tailing-based mortars both at 28 days and 120 days of curing. Especially in
544 the case of 120 days cured samples, the vibration bands corresponding to Si-O-Si/Al symmetric
545 stretching were found to be comparatively broader and intense than that of control mortar. This
546 is mainly ascribed to the formation of additional CSH and CASH phases through the effective
547 hydration reactions facilitated by the improved particle packing achieved from the fine nature
548 of mine tailings⁶³. The finer fractions of mine tailings also contribute in the additional
549 formation of CSH and CASH gels through secondary hydration reactions. The additional peak
550 found at the wavenumber 738.95 cm^{-1} could be due to the formation of additional C-S-H/CASH
551 phases by the intrusion of IOT in mortar. All these observations are found to be in good
552 alignment with the XRD analysis data.

553 By observing the color scales, there is also a considerable shift of Si-O-Si/Al symmetric
554 stretching bands towards the lower wavenumber by the inclusion of mine tailings. This fact
555 gives the information that stronger bonds were developed by the formation of additional CSH
556 and CASH phases in mine tailing based mortars⁶⁷⁻⁷⁰. It is also important to notice that there is
557 no significant shift observed in the wavenumbers associated to the Si-O-Si/Al symmetric
558 stretching by the inclusion of mine tailing by the end of 28 days of curing. This could be due
559 to the negligible amount or absence of secondary hydration reactions at the early stages of
560 curing by the delayed hydration activity caused by the heavy metals^{55,56}.

561 However, there also observed a drastic reduction in the intensities of vibration bands
562 corresponding to O-C-O asymmetric stretching and O-C-O bending with the usage of mine
563 tailings in both 28 days 120 days cured mortar samples. This behavior is attributed to the
564 increased compactness in the mine tailing based mortars owing to accelerated hydration
565 activity that leads to the formation of additional CSH and CASH gels, which acts as
566 carbonation barrier⁷¹. This feature helped the mine tailing-based mortars to have an advanced
567 quality microstructure even though the hydration reactions lagged in the early stages of curing.
568 These observations are in good alignment with the TGA and SEM studies.

Table 6: Details of the functional group assignment at different wavenumbers

Notation No	Age of curing	Wavenumber (cm ⁻¹) corresponding to different peaks								Functional group assignment
		CM	I10	I20	I30	C10	C20	C30	C40	
1	28	3305	3343	3385	3418	3372	3259	3531	3237	O-H asymmetric stretching (Due to bonded water in portlandite and free water)
	120	3513	3469	3427	3489	3342	3541	3621	3299	
2	28	1602	1631	1677	1648	1683	1634	1698	1613	O-H asymmetric bending (Due to free water)
	120	1651	1650	1648	1591	1601	1620	1643	1642	
3	28	1416	1464	1416	1439	1416	1412	1427	1412	O-C-O asymmetric stretching (Due to carbonates)
	120	1410	1412	1408	1437	1408	1401	1412	1422	
4	28	1086	1084	1096	1098	1094	1096	1092	1110	S-O stretching (Due to gypsum and ettringites)
	120	1087	1095	1081	1094	1115	1102	1086	1113	
5	28	-	993	-	998	-	-	-	-	Si-O-Si/Al asymmetric stretching (Due to silicates and aluminosilicate hydrates)
	120	1002	965	954	963	973	971	954	963	
6	28	872	874	872	879	870	874	885	-	O-C-O bending (Due to carbonates)
	120	874	874	874	875	879	883	875	872	
7	28	774	782	782	778	770	778	786	774	Si-O-Si/Al symmetric stretch (Due to silicates and aluminosilicate hydrates)
		716	714	-	716	-	-	-	-	
		692	689	699	-	-	693	689	-	
	120	778	767	773	784	790	786	765	763	
		-	-	-	738	-	-	-	-	
		686	685	691	679	681	689	-	695	
8	28	632	642	655	620	669	647	640	649	S-O bending (Due to gypsum and ettringites)
		-	-	-	-	624	-	-	624	
	120	634	610	618	616	628	645	653	653	
		-	-	-	-	-	-	626	624	

571 3.4.4 Scanning electron microscopy (SEM) and energy dispersive x-ray spectroscopy (EDS)

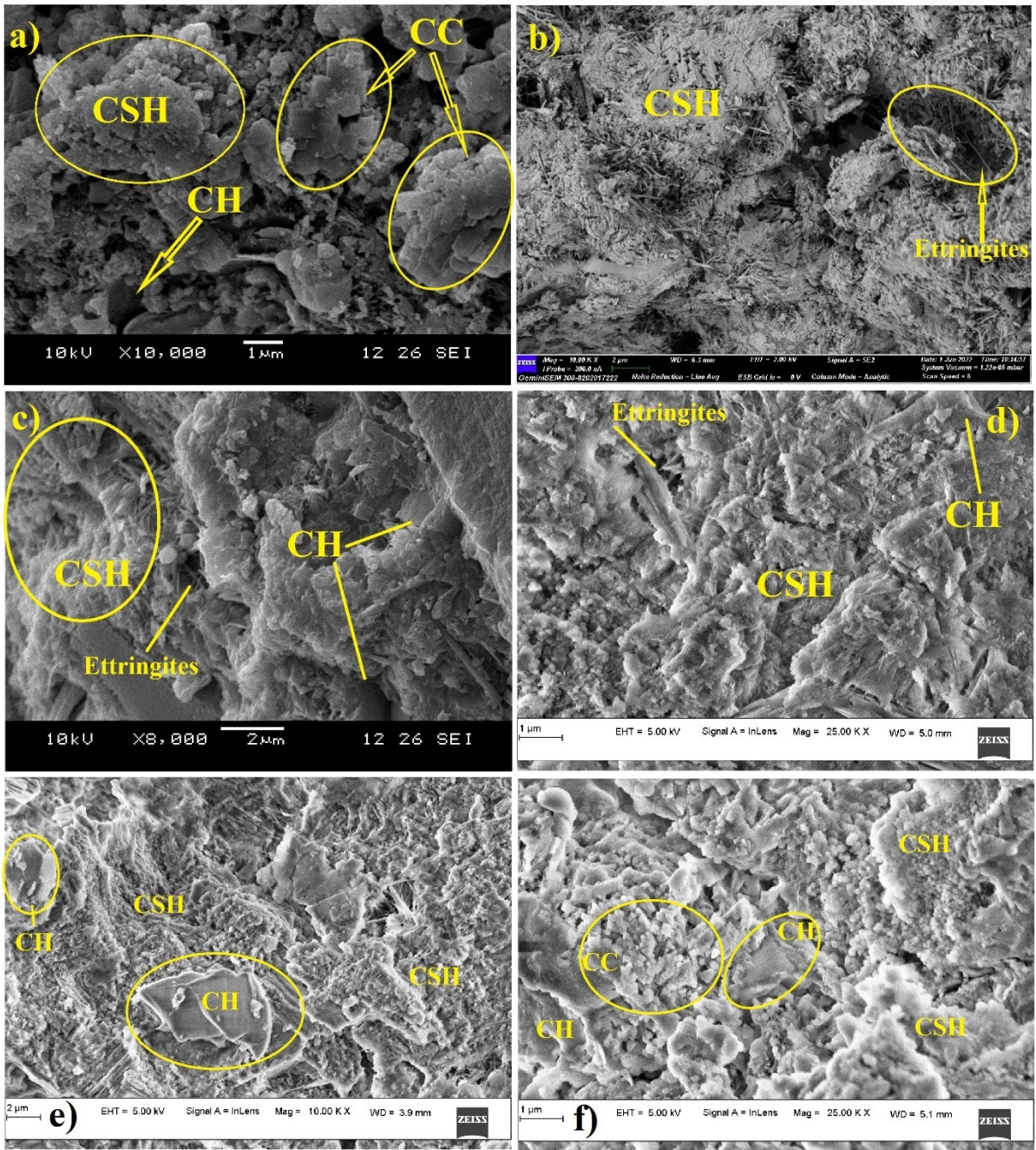
572 The morphology and the microstructure of hydrated samples of control, optimized IOT and
573 COT mortars (i.e., CM, I20 and C30) were studied using SEM at the curing age of 28 days and
574 120 days. Fig. 11 and Fig. 12 depicts the scanning electron micrographs of 28 days and 120
575 days cured mortar samples, respectively. The compounds in the micrographs were visually
576 identified based on their structure, shape, color and texture⁷². Further the elemental
577 composition of these mortar samples was investigated using area method in EDS technique.
578 The Ca/Si atomic ratio was calculated from the obtained results and tabulated in Table 7. It
579 was used to examine the extent of formation of major products related to hydration such as
580 calcium hydroxide (CH) and CSH. Numerous researches have claimed that the formation of
581 CSH through hydration process highly depends on the calcium (Ca) and silicate (Si) ions
582 present in the pore solution⁷²⁻⁷⁴. It has been reported that the cement matrix having lower Ca/Si
583 atomic ratio signifies to have a dense microstructure owing to the formation of strongly built
584 CSH network^{75,76}.

585 By observing the SEM micrographs of 28 days cured samples, it is possible to notice that the
586 mine tailings-based mortars developed additional portlandite, CSH and CASH phases from to
587 the accelerated hydration reactions facilitated by the superior particle packing, added calcium
588 oxide and iron oxide compounds from the finer fractions of mine tailings^{57,58,61,62}. The
589 increased Ca/Si ratio in mine tailing-based mortars is evident for the increased formation of
590 portlandite. However, the extensive formation of CSH and CASH phases in mine tailing-based
591 mortars helped in achieving a quality microstructure which is responsible for the improved
592 hardened properties.

593 Similar kind of observations can also be made in 120 days cured mortar. But the scanning
594 electron micrographs were found to have much more CSH and CASH phases due to the
595 succession of hydration reactions with the curing age. As depicted in the SEM micrographs,
596 the iron ore tailing based mortar showed globular, needled like and amorphous network of CSH
597 and CASH phases. Whereas, floccular, fibrous and plate like structured CSH and CASH phases
598 were formed in COT bases mortars. These diversely formed CSH and CASH phases beyond
599 28 days of curing in the mine tailing-based mortars are evident for the formation of additional
600 hydration products through secondary hydration reactions, facilitated by the reactive finer
601 fractions present in mine tailings. The reduction in Ca/Si atomic ratio in mine tailings mortar
602 at the end of 120 days curing was also significantly higher than control mortar due to the
603 effective utilization of CH for secondary hydration reactions. It is also important to notice that
604 the occurrence of higher volume of CC and ettringites can also be observed only in the SEM
605 images of control mortar. Whereas the mine tailing-based mortars illustrates less CC and
606 ettringites with ample amount of CSH and CASH.

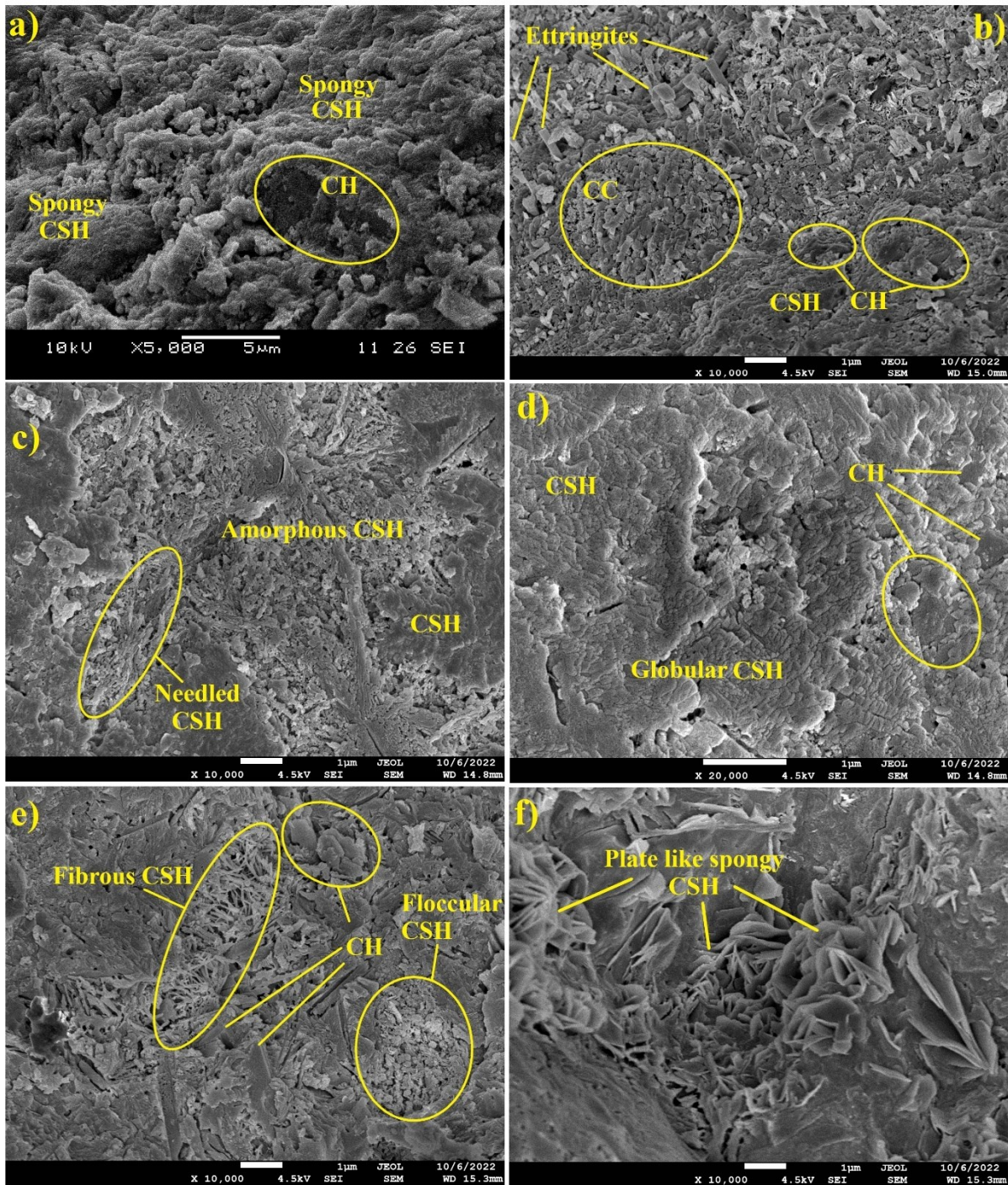
607 Thus, the additional hydration products formed from the primary and secondary hydration
608 reactions helps in densifying the mortar microstructure which steer the mine tailings-based
609 mortars to perform better than conventional mortar.

610



611

612 **Fig. 11. Scanning electron micrograph of control (CM) [in a), b)], I20 [in c), d)] and C30**
 613 **[in e), f)] mortar at 28 days of curing.**



614
615
616

Fig. 12. Scanning electron micrograph of control (CM) [in a), b)], I20 [in c), d)] and C30 [in e), f)] mortar at 120 days of curing.

617
618

Table 7: EDS elemental analysis of control, IOT and COT based mortars after a curing period of 28 days and 120 days

Element	Atomic weight (%) of 28 days cured mortar			Atomic weight (%) of 120 days cured mortar		
	CM	I20	C30	CM	I20	C30
C	15.43	14.04	15.38	10.00	12.05	17.05
Ca	21.20	20.97	20.96	9.85	15.70	10.90

O	54.79	55.69	55.48	60.60	50.90	51.05
Al	0.22	0.75	0.81	2.55	0.95	3.40
Si	7.475	6.9	6.28	4.05	6.60	5.50
Mg	0.16	0.14	0.18	-	0.25	0.05
S	0.28	0.28	0.28	10.65	3.35	0.75
Cl	0.03	0.05	0.02	-	-	-
K	0.11	0.08	0.13	0.30	0.60	1.80
Fe	0.3	1.1	0.48	1.75	1.55	0.10
Na	-	-	0.21	0.25	0.05	0.65
P	-	-	-	-	3.10	-
Mn	-	-	-	-	4.80	1.45
Ti	-	-	-	-	-	0.30
Zr	-	-	-	-	-	1.45
Ca/Si	2.83675	3.03913	3.33758	2.432	2.378	1.98198

619 4. Conclusions

620 After analyzing the experimental results in detail, the following conclusions can be drawn,

- 621 • The preliminary investigations on the mine tailings used in the present study revealed
622 that they have comparable properties with the river sand and hence could be used as an
623 alternative to river sand in developing the cementitious composites.
- 624 • The fresh properties such as workability and setting time reduce with the inclusion of
625 mine tailings as a replacement to river sand. The present study confirms the requirement
626 of superplasticizer and recommends desired volumes of superplasticizer (about 0-
627 0.45% by weight of binder) for achieving the required flow of 150 mm in different
628 mortars mixes.
- 629 • The improved particle packing by the inclusion of mine tailings till certain replacement
630 level of river sand helped in improving the hardened properties of mortar such as
631 compressive strength, water absorption and permeable porosity. However, the dry
632 density of mortars greatly relies on the specific gravity of tailings used.
- 633 • The outcome from the characterization studies such as lowering of Ca/Si ratio in EDS,
634 formation of additional CSH and CASH peaks in XRD, increased consumption of CH
635 observed in TGA and occurrence of broad, intensive Si-O-Si/Al stretching bands with
636 considerable shifts towards lower wavenumber in FTIR studies confirms the occurrence
637 of secondary hydration reactions by the finer fractions of mine tailings which helped in
638 improving the microstructure (can be observed through SEM images) and overall
639 performance (compressive strength, water absorption and permeable porosity) of the
640 cement mortar.
- 641 • The study concludes that, 20% of IOT and 30 % of COT by volume of river sand can
642 be considered as a potential replacement level for achieving superior performance in
643 cement mortar.
- 644 • By the end of 120 days of curing, I20 showed 20.53% higher compressive strength,
645 4.92% higher dry density, 1.20% lower water absorption and 2.56% lower permeable
646 porosity than the control mortar. Similarly, C30 showed 17.08% higher compressive

647 strength, 2.17% lower dry density, 1.78% lower water absorption and 3.13% lower
648 permeable porosity.

- 649 • Fineness of the material, angular particle shape along with the presence of higher
650 concentration of silica and alumina in the finer fractions of COT played a key role in
651 achieving the higher level of replacement for copper ore tailings in comparison with
652 iron ore tailings.
- 653 • The rough textured fine particles of IOT along with higher concentration of Fe₂O₃,
654 helped in achieving higher density and better performance in iron ore tailing based
655 mortar.

656 Declaration of Competing Interest

657 The authors declare that they have no conflict of interest, financial interests or personal
658 relationships that could have appeared to influence the research work reported in this article.

659 References

- 660 1. J. R. Owen, D. Kemp, É. Lébre, K. Svobodova, and G. P. Murillo, “Catastrophic tailings
661 dam failures and disaster risk disclosure,” *International journal of disaster risk
662 reduction* 42, (January 2020):101361, <https://doi.org/10.1016/j.ijdr.2019.101361>.
- 663 2. F. F. Carmo, A. O. Lanchotti, and L. H. Y. Kamino, “Mining waste challenges:
664 environmental risks of gigatons of mud, dust and sediment in megadiverse regions in
665 Brazil,” *Sustainability* 12, no. 20 (October 2020): 8466,
666 <https://doi.org/10.3390/su12208466>.
- 667 3. W. Pytel, “Risk assessment of mine tailings/waste surface ponds,” in *Mine Waste 2010:
668 Proceedings of the First International Seminar on the Reduction of Risk in the
669 Management of Tailings and Mine Waste*, (2010): 229–242,
670 https://doi.org/10.36487/ACG_rep/1008_20_Pytel.
- 671 4. Z. Róžański, “Fire hazard in coal waste dumps—selected aspects of the environmental
672 impact,” in *IOP Conference Series: Earth and Environmental Science* 174, no. 1 (2018):
673 12013, [DOI 10.1088/1755-1315/174/1/012013](https://doi.org/10.1088/1755-1315/174/1/012013).
- 674 [5] C. Cacciuttolo, D. Cano, and M. Custodio, “Socio-Environmental Risks Linked with
675 Mine Tailings Chemical Composition: Promoting Responsible and Safe Mine Tailings
676 Management Considering Copper and Gold Mining Experiences from Chile and Peru,”
677 *Toxics*, vol. 11, no. 5, p. 462, 2023.
- 678 [6] D. P. Mohapatra and D. M. Kirpalani, “Process effluents and mine tailings: sources,
679 effects and management and role of nanotechnology,” *Nanotechnology for
680 Environmental Engineering*, vol. 2, no. 1, pp. 1–12, 2017, doi: 10.1007/s41204-016-
681 0011-6.
- 682 [7] C. Falagán, B. M. Grail, and D. B. Johnson, “New approaches for extracting and
683 recovering metals from mine tailings,” *Minerals Engineering*, vol. 106, pp. 71–78, 2017,
684 doi: 10.1016/j.mineng.2016.10.008.
- 685 [8] H. R. Watling, “Review of biohydrometallurgical metals extraction from polymetallic
686 mineral resources,” *Minerals*, vol. 5, no. 1, pp. 1–60, 2014, doi: 10.3390/min5010001.
- 687 [9] Geological Survey of India, “Government of India Ministry of Mines REPORT OF THE

- 688 WORKING GROUP ON (OTHER THAN COAL & LIGNITE) THE TWELFTH
689 FIVE YEAR PLAN SUB GROUP – I,” 2011.
- 690 [10] Indian Bureau of Mines, “Part-I: General Reviews,” *Indian Mineral Yearbook 2012*, vol.
691 2012, no. 0712, pp. 1–7, 2014.
- 692 [11] C. Sudha, A. K. Kottuppillil, P. T. Ravichandran, and K. Divya Krishnan, “Study on
693 Mechanical Properties of Concrete with Manufactured Sand and Bagasse Ash,” *Indian
694 Journal of Science and Technology*, vol. 9, no. 34, 2016, doi:
695 10.17485/ijst/2016/v9i34/95867.
- 696 [12] M. Pilegis, D. Gardner, and R. Lark, “An investigation into the use of manufactured
697 sand as a 100% replacement for fine aggregate in concrete,” *Materials*, vol. 9, no. 6,
698 2016, doi: 10.3390/ma9060440.
- 699 [13] W. H. Langer and B. F. Arbogast, “Environmental impacts of mining natural aggregate,”
700 *Deposit and geoenvironmental models for resource exploitation and environmental
701 security*, pp. 151–169, 2002.
- 702 [14] L. H. Anh *et al.*, “Life Cycle Assessment of River Sand and Aggregates Alternatives in
703 Concrete,” *Materials*, vol. 16, no. 5, p. 2064, 2023.
- 704 [15] F. Muleya, B. Mulenga, S. L. Zulu, S. Nwaubani, C. K. Tembo, and H. Mushota,
705 “Investigating the suitability and cost-benefit of copper tailings as partial replacement
706 of sand in concrete in Zambia: an exploratory study,” *Journal of Engineering, Design
707 and Technology*, vol. 19, no. 4, pp. 828–849, 2021.
- 708 [16] A. Bandopadhyay, R. Kumar, and P. Ramachandrarao, “Clean technologies for
709 metallurgical industries (EWM-2002),” 2002.
- 710 [17] S. Zhang *et al.*, “Current situation and comprehensive utilization of iron ore tailing
711 resources,” *Journal of mining Science*, vol. 42, pp. 403–408, 2006.
- 712 [18] A. U. Shettima, Y. Ahmad, M. W. Hussin, N. Z. Muhammad, and O. E. Babatude,
713 “Strength and microstructure of concrete with iron ore tailings as replacement for river
714 sand,” in *E3S web of conferences*, 2018, vol. 34, p. 1003.
- 715 [19] A. U. Shettima, M. W. Hussin, Y. Ahmad, and J. Mirza, “Evaluation of iron ore tailings
716 as replacement for fine aggregate in concrete,” *Construction and Building Materials*,
717 vol. 120, pp. 72–79, 2016.
- 718 [20] S. Oritola, A. L. Saleh, and A. R. Mohd Sam, “Performance of iron ore tailings as partial
719 replacement for sand in concrete,” *Applied Mechanics and Materials*, vol. 735, pp. 122–
720 127, 2015.
- 721 [21] Z. Tian, Z. Zhao, C. Dai, and S. Liu, “Experimental study on the properties of concrete
722 mixed with iron ore tailings,” *Advances in Materials Science and Engineering*, vol.
723 2016, 2016.
- 724 [22] Z. Zhang, Z. Zhang, S. Yin, and L. Yu, “Utilization of iron tailings sand as an
725 environmentally friendly alternative to natural river sand in high-strength concrete:
726 shrinkage characterization and mitigation strategies,” *Materials*, vol. 13, no. 24, p. 5614,
727 2020.
- 728 [23] R. C. Gupta, P. Mehra, and B. S. Thomas, “Utilization of copper tailing in developing
729 sustainable and durable concrete,” *Journal of Materials in Civil Engineering*, vol. 29,

- 730 no. 5, p. 4016274, 2017.
- 731 [24] B. S. Thomas, A. Damare, and R. C. Gupta, "Strength and durability characteristics of
732 copper tailing concrete," *Construction and Building Materials*, vol. 48, pp. 894–900,
733 2013.
- 734 [25] A. M. Bhoi, Y. D. Patil, H. S. Patil, and M. P. Kadam, "Feasibility assessment of
735 incorporating copper slag as a sand substitute to attain sustainable production
736 perspective in concrete," *Advances in Materials Science and Engineering*, vol. 2018, pp.
737 1–11, 2018.
- 738 [26] K. S. Al-Jabri, M. Hisada, A. H. Al-Saidy, and S. K. Al-Oraimi, "Performance of high
739 strength concrete made with copper slag as a fine aggregate," *Construction and building
740 materials*, vol. 23, no. 6, pp. 2132–2140, 2009.
- 741 [27] ASTM C150/C150M-22, "Standard Specification for Portland Cement," *ASTM
742 C150/C150M-22*, no. Reapproved 1999, pp. 1–9, 2003, doi: 10.1520/C0150.
- 743 [28] BIS, "Indian Standard Drinking Water Specification (Second Revision)," *Bureau of
744 Indian Standards*, vol. IS 10500, no. May, pp. 1–11, 2012, [Online]. Available:
745 <http://cgwb.gov.in/Documents/WQ-standards.pdf>
- 746 [29] BIS (Bureau of Indian Standards), "Coarse and Fine Aggregate for Concrete-
747 Specification IS 383:2016," *New Delhi, India: BIS*. pp. 1–21, 2016.
- 748 [30] Bureau of Indian Standards, "IS 2645: 2003 Integral Waterproofing Compounds for
749 Cement Mortar and Concrete -Specification," p. 16, 2003.
- 750 [31] I. 9103-1999, "Indian standard concrete admixtures-specifications." Bureau of Indian
751 Standards New Delhi.
- 752 [32] X. Tian, H. Zhang, T. Zhang, and C. A. Fernández, "Alkali-activated copper tailings-
753 based pastes: Compressive strength and microstructural characterization," *Journal of
754 Materials Research and Technology*, vol. 9, no. 3, pp. 6557–6567, 2020.
- 755 [33] L. Cui, L. Wang, Y. Xu, X. Lou, and H. Wang, "Feasibility Evaluation of Replacing
756 River Sand with Copper Tailings," *Sustainability*, vol. 13, no. 14, p. 7575, 2021.
- 757 [34] B. S. Jain, "Green Geopolymers with Copper Tailings for Marine Environments," pp.
758 1–8, 2022.
- 759 [35] B. S. EN, "1052-3. BS EN 196-1: Methods of testing cement," *Determination of
760 strength*, 2005.
- 761 [36] B. S. EN, "1015-3. Methods of Test for Mortar for Masonry—Part 3: Determination of
762 Consistence of Fresh Mortar (by Flow Table)," *NBN, Bureau for Standardisation:
763 Brussels, Belgium*, 1999.
- 764 [37] ASTM C403, "ASTM C 403, 'Standard Test Method for Time of Setting of Concrete
765 Mixtures by Penetration Resistance,' ASTM C 403-95, Annual Book of ASTM
766 Standards, American Society for Testing and Materials, Pennsylvania, 1998. ASTM,"
767 pp. 1–7, 2005, doi: 10.1520/C0403.
- 768 [38] I. S. IS, "Indian Standard Is: 4031 (Part 6)-1988 (Reaffirmed 2005). Methods Of
769 Physical Tests For Hydraulic Cement Part 6 Determination Of Compressive Strength Of
770 Hydraulic Cement Other Than Masonry Cement (First Revision)," 2005.

- 771 [39] B. S. EN, “1015-10, 1999. Methods of test for mortar for masonry—Part 10:
772 Determination of dry bulk density of hardened mortar,” *Br. Stand.[29] BS EN*, pp. 1011–
773 1015, 1999.
- 774 [40] M. Safiuddin and N. Hearn, “Comparison of ASTM saturation techniques for measuring
775 the permeable porosity of concrete,” *Cement and Concrete Research*, vol. 35, no. 5, pp.
776 1008–1013, 2005.
- 777 [41] B. B. Das, D. N. Singh, and S. P. Pandey, “A comparative study for determining pore
778 volume of concrete,” *Indian Concrete Journal*, vol. 84, no. 12, p. 7, 2010.
- 779 [42] A. ASTM, “Standard test method for dry and wet bulk density, water absorption, and
780 apparent porosity of thin sections of glass-fiber reinforced concrete,” *ASTM C948-81*,
781 2016.
- 782 [43] K. Snehal, B. B. Das, and M. Akanksha, “Early age, hydration, mechanical and
783 microstructure properties of nano-silica blended cementitious composites,”
784 *Construction and Building Materials*, vol. 233, p. 117212, 2020.
- 785 [44] J. Zhang and G. W. Scherer, “Comparison of methods for arresting hydration of
786 cement,” *Cement and Concrete Research*, vol. 41, no. 10, pp. 1024–1036, 2011.
- 787 [45] J. W. Bullard *et al.*, “Mechanisms of cement hydration,” *Cement and concrete research*,
788 vol. 41, no. 12, pp. 1208–1223, 2011.
- 789 [46] R. Yu, P. Spiesz, and H. J. H. Brouwers, “Effect of nano-silica on the hydration and
790 microstructure development of Ultra-High Performance Concrete (UHPC) with a low
791 binder amount,” *Construction and Building Materials*, vol. 65, pp. 140–150, 2014.
- 792 [47] L. Soriano, J. Monzó, M. Bonilla, M. M. Tashima, J. Payá, and M. V Borrachero, “Effect
793 of pozzolans on the hydration process of Portland cement cured at low temperatures,”
794 *Cement and Concrete Composites*, vol. 42, pp. 41–48, 2013.
- 795 [48] L. P. Singh, A. Goel, S. K. Bhattacharyya, U. Sharma, and G. Mishra, “Hydration studies
796 of cementitious material using silica nanoparticles,” *Journal of Advanced Concrete
797 Technology*, vol. 13, no. 7, pp. 345–354, 2015.
- 798 [49] K. Snehal and B. B. Das, “Acid, alkali and chloride resistance of binary, ternary and
799 quaternary blended cementitious mortar integrated with nano-silica particles,” *Cement
800 and Concrete Composites*, vol. 123, p. 104214, 2021.
- 801 [50] K. Snehal and B. B. Das, “Pozzolanic reactivity and drying shrinkage characteristics of
802 optimized blended cementitious composites comprising of Nano-Silica particles,”
803 *Construction and Building Materials*, vol. 316, p. 125796, 2022.
- 804 [51] K. Snehal, B. B. Das, and S. Barbhuiya, “Influence of aggressive exposure on the
805 degradation of nano-silica admixed cementitious mortar integrated with phase change
806 materials,” *Construction and Building Materials*, vol. 335, p. 127467, 2022.
- 807 [52] K. Snehal and B. B. Das, “Effect of phase-change materials on the hydration and
808 mineralogy of cement mortar,” *Proceedings of the institution of civil engineers-
809 construction materials*, pp. 1–11, 2020.
- 810 [53] D. R. Dinger and J. E. Funk, “Particle-packing phenomena and their application in
811 materials processing,” *Mrs Bulletin*, vol. 22, no. 12, pp. 19–23, 1997.

- 812 [54] C. Emma and S. Distribution, “EMMA User Guide Version 1 User Guide Elkem
813 Materials Mixture Analyser-EMMA”, [Online]. Available: www.elkem.com
- 814 [55] M. Gou, L. Zhou, and N. W. Y. Then, “Utilization of tailings in cement and concrete: A
815 review,” *Science and Engineering of Composite Materials*, vol. 26, no. 1, pp. 449–464,
816 2019.
- 817 [56] D. Boakye and H. Uzoegbo, “Assessment of concrete with pulverized copper slag as
818 partial replacement of cement,” *Concr. Beton*, vol. 139, pp. 14–17, 2014.
- 819 [57] M. Yellishetty, V. Karpe, E. H. Reddy, K. N. Subhash, and P. G. Ranjith, “Reuse of iron
820 ore mineral wastes in civil engineering constructions: A case study,” *Resources,
821 Conservation and Recycling*, vol. 52, no. 11, pp. 1283–1289, 2008.
- 822 [58] J. R. Conner, *Chemical fixation and solidification of hazardous wastes*. Van Nostrand
823 Reinhold New York, 1990.
- 824 [59] F. S. Hashem, M. S. Amin, and E. E. Hekal, “Stabilization of Cu (II) wastes by C3S
825 hydrated matrix,” *Construction and Building Materials*, vol. 25, no. 8, pp. 3278–3282,
826 2011.
- 827 [60] M. F. M. Zain, M. N. Islam, S. S. Radin, and S. G. Yap, “Cement-based solidification
828 for the safe disposal of blasted copper slag,” *Cement and Concrete Composites*, vol. 26,
829 no. 7, pp. 845–851, 2004.
- 830 [61] H. Zhang, S. Mu, J. Cai, J. Liu, and J. Hong, “The role of iron in cement hydration
831 process: From perspective of chemical admixture,” *Thermochimica Acta*, vol. 722, p.
832 179457, 2023.
- 833 [62] M. A. Largeau, R. Mutuku, and J. Thuo, “Effect of Iron Powder (Fe_2O_3) on Strength,
834 Workability, and Porosity of the Binary Blended Concrete,” *Open Journal of Civil
835 Engineering*, vol. 8, no. 04, p. 411, 2018.
- 836 [63] Y. Yang, L. Chen, X. Sun, and Y. Mao, “Preparation of micro-iron ore tailings by wet-
837 grinding and its application in sulphoaluminate cement,” *Journal of Renewable
838 Materials*, vol. 10, no. 4, p. 1007, 2022.
- 839 [64] E. Kapeluszna, Ł. Kotwica, and W. Nocuń-Wczelik, “Comparison of the effect of
840 ground waste expanded perlite and silica fume on the hydration of cements with various
841 tricalcium aluminate content – Comprehensive analysis,” *Construction and Building
842 Materials*, vol. 303, no. July, 2021, doi: 10.1016/j.conbuildmat.2021.124434.
- 843 [65] M. Hu, X. Zhu, and F. Long, “Alkali-activated fly ash-based geopolymers with zeolite
844 or bentonite as additives,” *Cement and Concrete Composites*, vol. 31, no. 10, pp. 762–
845 768, 2009.
- 846 [66] H. Biricik and N. Sarier, “Comparative study of the characteristics of nano silica-, silica
847 fume-and fly ash-incorporated cement mortars,” *Materials Research*, vol. 17, pp. 570–
848 582, 2014.
- 849 [67] K. Snehal, B. B. Das, and S. Barbhuiya, “Synergistic effect of nano silica on carbonation
850 resistance of multi-blended cementitious mortar,” *Cement and Concrete Composites*,
851 vol. 141, p. 105125, 2023.
- 852 [68] L. Yu, Z. Zhang, X. Huang, B. Jiao, and D. Li, “Enhancement experiment on
853 cementitious activity of copper-mine tailings in a geopolymer system,” *Fibers*, vol. 5,

- 854 no. 4, p. 47, 2017.
- 855 [69] S. Ahmari, K. Parameswaran, and L. Zhang, “Alkali activation of copper mine tailings
856 and low-calcium flash-furnace copper smelter slag,” *Journal of materials in civil*
857 *engineering*, vol. 27, no. 6, p. 4014193, 2015.
- 858 [70] I. C. Ferreira *et al.*, “Reuse of iron ore tailings for production of metakaolin-based
859 geopolymers,” *Journal of Materials Research and Technology*, vol. 18, pp. 4194–4200,
860 2022.
- 861 [71] A. Jagadisha, K. B. Rao, G. Nayak, and M. Kamath, “Influence of nano-silica on the
862 microstructural and mechanical properties of high-performance concrete of containing
863 EAF aggregate and processed quarry dust,” *Construction and Building Materials*, vol.
864 304, p. 124392, 2021.
- 865 [72] A. Y. Patil *et al.*, “Multi-scale study on mechanical property and strength of new green
866 sand (Poly Lactic Acid) as replacement of fine aggregate in concrete mix,” *Symmetry*,
867 vol. 12, no. 11, p. 1823, 2020.
- 868 [73] Q. Hu *et al.*, “Direct three-dimensional observation of the microstructure and chemistry
869 of C3S hydration,” *Cement and concrete research*, vol. 88, pp. 157–169, 2016.
- 870 [74] E. P. Sumukh, S. K. Goudar, and B. B. Das, “Predicting the Service Life of Reinforced
871 Concrete by Incorporating the Experimentally Determined Properties of Steel–Concrete
872 Interface and Corrosion BT - Recent Trends in Civil Engineering,” 2021, pp. 399–417.
- 873 [75] E. P. Sumukh, S. K. Goudar, and B. B. Das, *A Review on the Properties of Steel-*
874 *Concrete Interface and Characterization Methods*, vol. 78. 2021. doi: 10.1007/978-981-
875 15-5001-0_15.
- 876 [76] S. K. Goudar, E. P. Sumukh, and B. B. Das, “Influence Of Marine Environment
877 Exposure On The Engineering Properties Of Steel-concrete Interface,” *The Open Civil*
878 *Engineering Journal*, vol. 16, no. 1, 2022.
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883 **Figure Captions:**

884 **Fig. 1. a) Particle size distribution of various materials used, b) Mineralogical composition**
885 **of fine aggregates used**

886 **Fig. 2. a) Mine tailing, b) scanning electron micrograph at 500X, c) scanning electron**
887 **micrograph at 10,000X**

888 **Fig. 3. Particle packing curves of a) IOT based mortars and b) COT based mortars**
889 **determined using modified Andreasen and Andersen particle packing model**

890 **Fig. 4. a) Flow characteristics of a) IOT based mortar; b) COT based mortar**

891 **Fig. 5. a) Compressive strength, c) Water absorption, e) permeable porosity of IOT based**
892 **mortars, and b) Compressive strength, d) Water absorption, f) Permeable porosity of**
893 **COT based mortars**

894 **Fig. 6. XRD pattern of optimized mortars at the curing age of a) 28 days and b) 120 days**

895 **Fig. 7. TG and DTG curves of a) IOT based mortar after 28 days of curing, b) IOT based**
896 **mortar after 120 days of curing, c) COT based mortar after 28 days of curing and d) COT**
897 **based mortar after 120 days of curing**

898 **Fig. 8. Phase compounds formed in IOT based mortar [(a) CH; (c) CC] and COT based**
899 **mortar [(b) CH; (d) CC] after 28 days and 120 days of curing**

900 **Fig. 9. FTIR spectra of control and IOT based mortar samples at a) 28 days of curing and**
901 **b) 120 days of curing**

902 **Fig. 10. FTIR spectra of control and COT based mortar samples at a) 28 days of curing**
903 **and b) 120 days of curing**

904 **Fig. 11. Scanning electron micrograph of control (CM) [in a), b)], I20 [in c), d)] and C30**
905 **[in e), f)] mortar at 28 days of curing.**

906 **Fig. 12. Scanning electron micrograph of control (CM) [in a), b)], I20 [in c), d)] and C30**
907 **[in e), f)] mortar at 120 days of curing.**

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